

# Chemistry Study Guide

# Periodic Table of the Elements

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<sup>3</sup> <sub>3</sub> Li	<sup>4</sup> <sub>4</sub> Be											<sup>5</sup> <sub>5</sub> B	<sup>6</sup> <sub>6</sub> C	<sup>7</sup> <sub>7</sub> N	<sup>8</sup> <sub>8</sub> O	<sup>9</sup> <sub>9</sub> F	<sup>10</sup> <sub>10</sub> Ne																															
<sup>11</sup> <sub>11</sub> Na	<sup>12</sup> <sub>12</sub> Mg											<sup>13</sup> <sub>13</sub> Al	<sup>14</sup> <sub>14</sub> Si	<sup>15</sup> <sub>15</sub> P	<sup>16</sup> <sub>16</sub> S	<sup>17</sup> <sub>17</sub> Cl	<sup>18</sup> <sub>18</sub> Ar																															
<sup>19</sup> <sub>19</sub> K	<sup>20</sup> <sub>20</sub> Ca	<sup>21</sup> <sub>21</sub> Sc	<sup>22</sup> <sub>22</sub> Ti	<sup>23</sup> <sub>23</sub> V	<sup>24</sup> <sub>24</sub> Cr	<sup>25</sup> <sub>25</sub> Mn	<sup>26</sup> <sub>26</sub> Fe	<sup>27</sup> <sub>27</sub> Co	<sup>28</sup> <sub>28</sub> Ni	<sup>29</sup> <sub>29</sub> Cu	<sup>30</sup> <sub>30</sub> Zn	<sup>31</sup> <sub>31</sub> Ga	<sup>32</sup> <sub>32</sub> Ge	<sup>33</sup> <sub>33</sub> As	<sup>34</sup> <sub>34</sub> Se	<sup>35</sup> <sub>35</sub> Br	<sup>36</sup> <sub>36</sub> Kr																															
<sup>37</sup> <sub>37</sub> Rb	<sup>38</sup> <sub>38</sub> Sr	<sup>39</sup> <sub>39</sub> Y	<sup>40</sup> <sub>40</sub> Zr	<sup>41</sup> <sub>41</sub> Nb	<sup>42</sup> <sub>42</sub> Mo	<sup>43</sup> <sub>43</sub> Tc	<sup>44</sup> <sub>44</sub> Ru	<sup>45</sup> <sub>45</sub> Rh	<sup>46</sup> <sub>46</sub> Pd	<sup>47</sup> <sub>47</sub> Ag	<sup>48</sup> <sub>48</sub> Cd	<sup>49</sup> <sub>49</sub> In	<sup>50</sup> <sub>50</sub> Sn	<sup>51</sup> <sub>51</sub> Sb	<sup>52</sup> <sub>52</sub> Te	<sup>53</sup> <sub>53</sub> I	<sup>54</sup> <sub>54</sub> Xe																															
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<sup>57</sup> <sub>57</sub> La	<sup>58</sup> <sub>58</sub> Ce	<sup>59</sup> <sub>59</sub> Pr	<sup>60</sup> <sub>60</sub> Nd	<sup>61</sup> <sub>61</sub> Pm	<sup>62</sup> <sub>62</sub> Sm	<sup>63</sup> <sub>63</sub> Eu	<sup>64</sup> <sub>64</sub> Gd	<sup>65</sup> <sub>65</sub> Tb	<sup>66</sup> <sub>66</sub> Dy	<sup>67</sup> <sub>67</sub> Ho	<sup>68</sup> <sub>68</sub> Er	<sup>69</sup> <sub>69</sub> Tm	<sup>70</sup> <sub>70</sub> Yb	<sup>71</sup> <sub>71</sub> Lu																																		
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## Objective:

To understand the *structure of an atom*, and the *nature + properties* of subatomic particles 🎯

## Matter:

Anything that has **mass** and **occupies space** ✖️🌐

Examples: air, gases, liquids, solids!

## Atoms:

Atoms are the **smallest, indivisible particle** 🧪👉

- **Outer part** → *shell/orbit/energy level* → carries the electrons ( $e^-$ )
- **Inner part** → *nucleus* → houses the protons ( $p^+$ ) and neutrons ( $n$ )
- Outer & Inner part → together these are the **subatomic particles**

## Subatomic Particle Table:

Particle	Symbol	Mass (amu)	Charge	Location
Proton	$p^+$	1	+1	Nucleus
Neutron	$n$	1	0	Nucleus
Electron	$e^-$	$\sim 0$	-1	Orbit/Shell

# Section 2 nuclear Chemistry Study Guide Key

**Susan McMurry**



## Section 2nuclear Chemistry Study Guide Key:

**Study Guide with Answers to Selected Problems** H. Stephen Stoker, Danny V. White, 2003-04 This supplement includes for each chapter a brief overview activities and practice problems to reinforce skills and a practice test The answers section includes answers for all odd numbered end of chapter exercises

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*Study Guide, Answer Book to Accompany Organic Chemistry* Alan S. Wingrove, Robert L. Caret, 1981

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*Study Guide and Partial Solutions Manual, Fundamentals of General, Organic, and Biological Chemistry* Susan McMurry, 1996 This internationally acclaimed detective series is just the thing for lovers of those Number One Ladies looking for a darker more realistic view of Botswana Sue Baker Publishing News

*Study Guide and Solutions Manual for McMurry's Organic Chemistry, Fifth Edition* Susan McMurry, 2000 Provides answers and explanations to all in text and end of chapter exercises Also includes summaries of name reactions functional group synthesis and reactions lists of reagents and abbreviations and articles on topics ranging from infrared absorption frequencies to the Nobel Price winners in Chemistry This edition now includes all new artwork expanded in text problems summary quizzes approximately every three

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