

## ANSWERS TO STUDY GUIDE QUESTIONS

### **Romeo and Juliet** **Study Guide – Act I**

#### Prologue:

1. In what city does this play take place?

*The play takes place in the Italian city of Verona.*

2. Why are Romeo and Juliet called “Star-cross’d lovers?”

*Romeo and Juliet are considered star-crossed lovers because fate has brought them together, yet their families are at war with one another and they will never be able to be together; therefore, fate will also tear them apart.*

#### Act I, Scene I:

1. Who is fighting at the beginning of the first scene?

*In the beginning of Act I Scene I, a fight occurs between the Capulets and the Montagues. The fight is between Sampson (Capulet), Gregory (Capulet), Abraham (Montague), and Balthazar (Montague). Later in the scene Benvolio (Montague) and Tybalt (Capulet) fight.*

2. Who tries to break up the fighting?

*Benvolio tries to break up the fighting but ends up joining the fight against Tybalt.*

3. What threat does the Prince make to Lord Montague and Lord Capulet?

*The Prince is tired of the fighting and threatens to execute anyone that becomes involved in a fight in the streets.*

4. Benvolio and Montague describe the way Romeo has been acting. What do they have to say about him?

*Benvolio and Montague talk about how depressed Romeo seems to be the last few days. Romeo is refusing to talk to anyone walking morosely through the gardens crying. Montague describes him as a flower that refuses to open up.*

5. Why is Romeo so sad? Explain.

*Romeo is in love with a woman named Rosaline. Rosaline has sworn that she will never love a man and wishes to join a convent (to become a nun). Romeo is sad that she does not love him back. Romeo feels alone and tells Benvolio that he will never be in love with another woman.*

6. What is Benvolio’s advice to Romeo?

# Romeo And Juliet Study Guide Questions And Answers Act

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web this is a spectacular version of the flame tests experiment that can be used with chemists and non chemists alike it can be extended as an introduction to atomic spectra for post 16 students this experiment must be done as a demonstration it takes about ten minutes if all is prepared in advance

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web experiment sweeping the detector arm through will now record a spectrum of the light from the discharge lamp try different aperture and slit sizes and adjusting the lens locations to record as many of the spectral peaks as possible dim spectral peaks will require careful tuning of the apertures to observe

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web to observe and understand line emission spectra of atoms using gas discharge tubes to practice writing electron configurations for these and other elements electromagnetic radiation is energy in the form of waves waves are characterized by their wavelength  $\lambda$  and frequency  $\nu$

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web the bohr model named after danish physicist niels bohr of an atom has a small positively charged central nucleus and electrons orbiting in at specific fixed distances from the nucleus

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web objectives to perform flame tests of metal cations in order to observe their featured colors to perform calculations up determine the frequency and energy in the emitted photons to tell these results the types of electronic transitions occurring in are default

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web atomic emission spectra electrons in atoms normally occupy the lowest energy states possible such an atom is said to be in the ground state however electrons can be excited to high energy states when they absorb excess energy the excess energy can be provided by heat light or electrical discharge

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web fingerprints of the elements atomic spectra as dr matilsky discussed in his video lecture atomic spectra occur due to the fact that orbital radii of electrons and hence their energies are quantized at specific levels determined by the atomic number number of protons and ionization state number of electrons in any given element

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web in this experiment the student measures the features of a simple spectrum that of sodium an empirical relation is deduced for the frequencies of the lines and the position of unobserved lines is predicted