

19.2

HYDROGEN IONS AND ACIDITY

Section Review

Objectives

1. Calculate K_a and K_b for weak acids and bases, and pK_a and pK_b for weak acids and bases.
2. Calculate K_a and K_b for weak acids and bases, and pK_a and pK_b for weak acids and bases.
3. Calculate the pH of weak acid and base solutions.

Vocabulary

- | | | |
|-----------------|--------------------------------------|-----------------|
| acid | acid dissociation constant (K_a) | acid ionization |
| acidic solution | acidic solution | acidic solution |
| acidic solution | acidic solution | acidic solution |

Key Equations

- | | |
|---|--|
| $K_a = \frac{[H^+][A^-]}{[HA]}$ (Equation 19.1) | $K_b = \frac{[OH^-][HB^+]}{[B]}$ (Equation 19.2) |
|---|--|

Part A Completion

Write the chemical formula for each of the following acids, bases, and salts. Identify the acid, base, or salt.

1. Hydrochloric acid, HCl , $HClO_4$, HNO_3 , H_2SO_4

2. Acetic acid, CH_3COOH , CH_3COO^- , CH_3COO^-

3. Ammonia, NH_3 , NH_4^+ , NH_4^+

4. Hydrofluoric acid, HF , HF , HF

5. Hydrobromic acid, HBr , HBr , HBr

6. Hydroiodic acid, HI , HI , HI

7. Hydrocyanic acid, HCN , HCN , HCN

8. Hydroperfluoroboric acid, HF_3BO_3 , HF_3BO_3

9. Hydroperfluoroboric acid, HF_3BO_3 , HF_3BO_3

10. Hydroperfluoroboric acid, HF_3BO_3 , HF_3BO_3

11. Hydroperfluoroboric acid, HF_3BO_3 , HF_3BO_3

1. Hydrochloric acid
2. Acetic acid
3. Ammonia
4. Hydrofluoric acid
5. Hydrobromic acid
6. Hydroiodic acid
7. Hydrocyanic acid
8. Hydroperfluoroboric acid
9. Hydroperfluoroboric acid
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Maurice Bleifeld



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