

# AP Chemistry Final Study Guide

## Unit 1: Atomic Structure and Properties (7%–9%)

- At STP, one mole of gas always occupies 22.4 L
- **Mass Spectroscopy:**
  - Atoms have different isotopes (carbon-12, carbon-14, etc) and this will affect the average atomic mass
  - The most abundant isotope will have the greatest effect
- **Coulomb's Law:**
  - The greater the charge of the nucleus, the more energy is required to ionize an electron
  - The closer an electron is to the nucleus, the more energy is required to ionize an electron
- **Electromagnetic Radiation:**
  - When atoms absorb energy, electrons jump to higher energy levels
  - When electrons drop from higher to lower energy levels, atoms give off electromagnetic radiation energy
- **Kinetic Energy of Ionized Electrons:**
  - The faster an ejected electron is going, the more kinetic energy it has
  - Electrons originally further away from the nucleus require less energy to eject, so it will have more kinetic energy
- **Electron Configuration:**
  - When entering the d-block (transition metals), the energy level drops by 1 unless you are using shorthand notation
    - Nickel:  $1s^2 2s^2 2p^6 3s^2 3p^6 3d^8 4s^2$  (notice the d-block)
    - Nickel (with shorthand):  $[\text{Ar}] 4s^2 3d^8$
- **Law of Definite Proportions:**
  - A pure chemical compound broken down into elements always contains elements of a fixed ratio, independent of where and how it was created
- **Law of Multiple Proportions:**
  - When elements form compounds, the proportions of the elements in those chemical compounds can be expressed in small whole number ratios
- **Aufbau Principle:**
  - When building up the electron configuration of an atom, electrons are placed in orbitals, subshells, and shells in order of increasing energy

# Ap Chemistry Test Study Guide

**Brian Stewart**



## Ap Chemistry Test Study Guide:

**CliffsNotes AP Chemistry** Bobrow Test Preparation Services, 2009-02-09 The book itself contains chapter length subject reviews on every subject tested on the AP Chemistry exam as well as both sample multiple choice and free response questions at each chapter's end Two full length practice tests with detailed answer explanations are included in the book

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The Girls' Guide to the SAT Alexandra Freer, Princeton Review (Firm), 2003 It s scary enough that the SAT can make or break one s college admission chances but the fact that girls consistently score lower than boys makes it an even bigger hurdle The Girls Guide to the SAT helps young women understand and overcome the gender gap with specially focused tips and techniques for scoring higher

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Cracking the AP Chemistry Paul Foglino, Princeton Review (Firm), 2004. The fiercer the competition to get into college the more schools require that students prove themselves in other ways than SAT scores and grade point averages. The more expensive college educations become the more students take advantage of the opportunity to test out off-year college courses. Includes 2 sample tests with full explanations for all answers. The Princeton Review's proven score raising skills and techniques. Complete subject review of all the material likely to show up on the AP Chemistry exam. *Barron's ACT Study Guide Premium, 2023: 6 Practice Tests + Comprehensive Review + Online Practice* Brian Stewart, 2023-01-03. Always study with the most up to date prep. Look for ACT Study Guide Premium Prep 2024 6 Practice Tests Comprehensive Review Online Practice ISBN 9781506291543 on sale March 5 2024. Publisher's Note: Products purchased from third party sellers are not guaranteed by the publisher for quality authenticity or access to any online entities included with the product. *5 Steps to a 5: 500 AP Chemistry Questions to Know by Test Day, Third Edition* Mina Lebitz, 2019-02-14. 500 AP style questions with detailed answer explanations to prepare you for what you'll see on test day. *5 Steps to a 5 500 AP Chemistry Questions to Know by Test Day* gives you 500 practice questions that cover the most essential course material and help you work toward a 5 on the test. The questions parallel the format and degree of difficulty that you'll find on the actual AP exams and are accompanied by answers with comprehensive explanations. The questions in this book were written by expert AP teachers who know the exam inside and out so they closely reflect what you'll see when you sit for the AP Chemistry test. This valuable study guide features 500 AP style questions and answers. Detailed review explanations for right and wrong answers. Close simulations of the real AP exam. Updated material that reflects the latest AP exam. AP Chemistry For Dummies Peter J. Mikulecky, Michelle Rose Gilman, Kate Brutlag, 2008-11-13. A practical and hands on guide for learning the practical science of AP chemistry and preparing for the AP chem exam. *Gearing up for the AP Chemistry exam* AP Chemistry For



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