



A PERIODIC TABLE LOGIC PROBLEM

By Mark L. Brader, Ph.D., and James M. Hartman, Ph.D.

OBJECTIVES

- Analyze properties of elements, groups, and subgroups.
- Identify and analyze groups of the periodic table that are similar to one another.

INTRODUCTION

Periodic tables are important for the organization of the elements because they make predictions about their physical and chemical properties. The most common tables, those containing the elements in groups, are useful for predicting the properties of elements. These are a continuing and never-ending source.

The first periodic table was proposed by the Russian chemist Dmitri Mendeleev in 1869. He organized the elements into groups of two or more elements with similar properties.

Mendeleev's periodic table was based on the atomic weights of the elements. He arranged the elements in order of increasing atomic weight. The elements in the same group had similar properties.

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Bruce Reinhart

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