

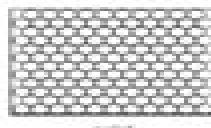
> Workbook answers

Chapter 1

Exercise 1.1

1 A solid has a **fixed mass** and **volume / shape**.
 A liquid has a **fixed mass** but its **shape changes to that of the container in which it is placed**.
 A gas has **no fixed shape or volume**. A gas **completely fills the container that it is in**.

2



solid



liquid



gas

3 a B
 b C
 c A
 d D

4 A freezing (solidification)

B melting

C condensation

D evaporation or boiling

5 a radon

b radon and nitrogen

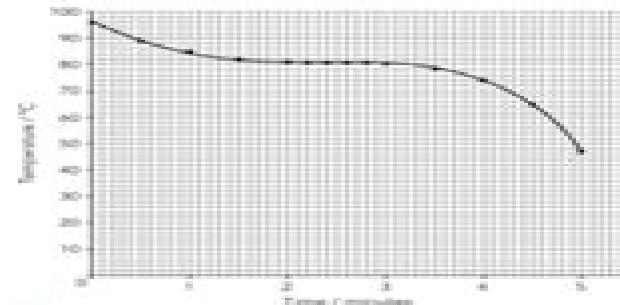
c nitrogen

d cobalt

e The sample of ethanoic acid is impure.
 The presence of impurities raises the boiling point of a substance.

Exercise 1.2

6



7 The student would need to use an oil bath (in place of the water bath) so that the higher temperature could be reached.

8 The substance is freezing (solidifying) / turning from liquid to solid.

9 The temperature stays constant because energy is being released as the substance solidifies / the molecules are giving out heat as they stop moving from place to place and become organised in a structured lattice arrangement / in the solid the molecules can only vibrate about fixed points / the heat released by the formation of new interactions (forces) between the particles keeps the temperature constant until all the substance is solid.

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