

PM SHRI KENDRIYA VI DYALAYA GACHI BOWLI, GPRA CAMPUS, HYD-32  
 PRACTICE PAPER 11 (2024- 25)  
 CHAPTER 04 CARBON AND ITS COMPOUND

SUBJECT: SCIENCE

MAX. MARKS : 40

CLASS : X

DURATION : 1½ hrs

**General Instructions:**

- All questions are compulsory.
- This question paper contains 20 questions divided into five Sections A, B, C, D and E.
- Section A comprises of 10 MCQs of 1 mark each. Section B comprises of 4 questions of 2 marks each. Section C comprises of 3 questions of 3 marks each. Section D comprises of 1 question of 5 marks each and Section E comprises of 2 Case Study Based Questions of 4 marks each.
- There is no overall choice.
- Use of Calculators is not permitted

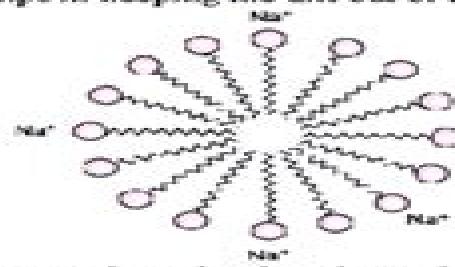
**SECTION - A**

Questions 1 to 10 carry 1 mark each.

- Ethanol reacts with sodium and forms two products. These are:
 

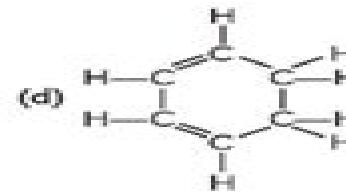
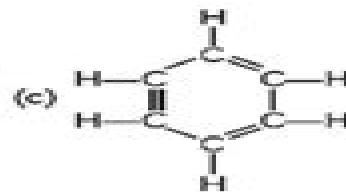
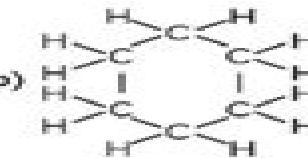
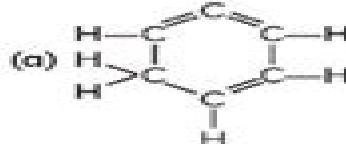
(a) sodium ethanoate and hydrogen	(b) sodium ethanoate and oxygen
(c) sodium ethoxide and hydrogen	(d) sodium ethoxide and oxygen
- A student conducts an activity where he burns methane in the presence of oxygen. What is likely to form?
 

(a) Water	(b) Carbon dioxide
(c) Carbon dioxide and water	(d) Carbon dioxide and oxygen
- A student studies that soap solution results in micelle formation which helps to remove dirt. It has a unique orientation which helps in keeping the dirt out of the water as shown in the image.



The student was curious to know more about the cleansing action of the soap. What helps the dirt to rinse away?

- Suspension of the dirt in the micelles.
  - A collection of water molecules in the centre of the micelle.
  - Attraction between the ionic end and the dirt to remove it.
  - Mixing of the soap molecules along with the dirt to make it heavier.
- The structural formula of benzene is:



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