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web aug 15 2020 the equilibrium formulated in scheme 1 between hydrogen ion and the corresponding acid and base can be called a simple acid base equilibrium by mixing two simple systems a double acid base system and an acid base equilibrium result that can always be formulated as follows acid<sub>1</sub> base<sub>2</sub> acid<sub>2</sub> base<sub>1</sub>

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web the a ion is a base because it is accepting a proton from the h<sub>3</sub>o the reversible reaction contains two acids and two bases we think of them in pairs called conjugate pairs when the acid ha loses a proton it forms a base a when the base a accepts a proton back again it obviously reforms the acid ha

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electrons instead of proton transfer and specifically stated that an acid is a species that accepts

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