

# Cognitive Ability Test Challenge Cards



## Cognitive Ability Test

12. Look at the shapes in the boxes below. Carefully look at the pattern of the shapes going across and up and down. Can you see the pattern? Considering this pattern, which of the shapes below fits in the empty box?



## Cognitive Ability Test

21. What is the missing number from the equation below?

$$(2 \times 3) + 10 = (10 \div 2) + 5$$

- a. 1
- b. 2
- c. 3
- d. 4
- e. 5

## Cognitive Ability Test

2. The boy slides down the \_\_\_\_\_

- a. donkey
- b. fruit tree
- c. ladder
- d. slide



## Cognitive Ability Test

15. Look at the shapes in the boxes below. Carefully look at the pattern of the shapes going across and up and down. Can you see the pattern? Considering this pattern, which of the shapes below fits in the empty box?



## Cognitive Ability Test

23. The numbers below follow a pattern. Can you see it? Follow the pattern to work out the missing number.

62 56 50 44 ?

- a. 20
- b. 32
- c. 34
- d. 36
- e. 38

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# Year 5 Cats Test Sample Questions

**Karen Lurie**



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web determine the lewis structure molecular geometry and polarity for molecules using lewis structures sketch molecular geometries on paper with wedges and dashes to represent their 3d shape explain how  $\pi$  bonding affects the ability of atoms to rotate about a bond relate molecular polarity to physical properties

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web jul 14 2012 this activity combines two simulations molecular shapes and molecular polarity to guide the students from drawing lewis structures to vsepr predicted geometries to predicting a molecule s polarity subject chemistry level undergrad



intro type homework lab duration 90 minutes answers included no

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web draw lewis structure use vsepr to determine molecular geometry determine bond polarity based on electronegativity differences determine molecular polarity based on bond dipoles molecular geometry for the following molecules complete this

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web two regions of electron density around a central atom in a molecule form a linear geometry three regions form a trigonal planar geometry four regions form a tetrahedral geometry five regions form a trigonal bipyramidal geometry and six regions form an octahedral geometry

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web molecules composed of covalently bonded atoms may also be polar or nonpolar for the molecule to be polar it must of course have polar bonds but the key factor for determining the polarity of a molecule is its shape if the polar bonds dipoles are symmetrical around the central atom they offset each other and the resulting molecule is

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web linear geometry all angles 180 degrees three regions of electron density around a central atom in a molecule form a trigonal planar geometry all angles 120 degrees four regions of electron density around a central atom in a molecule form a

tetrahedral geometry all angles 109.5 degrees

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web in this investigation you will examine a theory that chemists use to explain different aspects of chemical bonding valence shell electron pair repulsion vsepr theory attention will be given to how molecules are arranged in different shapes and how chemists can predict the geometry of a given molecule