

# Cognitive Ability Test

## Challenge Cards



### Cognitive Ability Test

12. The boy likes doing the \_\_\_\_\_.

- a. dancing
- b. fruit tree
- c. dancing
- d. slide



### Cognitive Ability Test

13. Look at the shapes in the box below. Carefully look at the pattern of the shapes going across and up and down. Can you see the pattern? Considering this pattern, which of the shapes below fits in the empty box?



### Cognitive Ability Test

14. What is the missing number from the equation below?

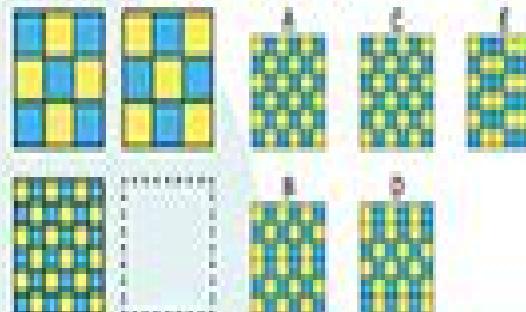
$$(7 + 1) \times 10 + (10 + 2) = 9$$

- a. 1
- b. 2
- c. 3
- d. 4
- e. 5



### Cognitive Ability Test

15. Look at the shapes in the box below. Carefully look at the pattern of the shapes going across and up and down. Can you see the pattern? Considering this pattern, which of the shapes below fits in the empty box?



$$62 \quad 56 \quad 50 \quad 44 \quad ?$$

- a. 39
- b. 37
- c. 34
- d. 36
- e. 38

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# Year 5 Cats Test Sample Questions

**Karen Lurie**

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web determine the lewis structure molecular geometry and polarity for molecules using lewis structures sketch molecular geometries on paper with wedges and dashes to represent their 3d shape explain how  $\pi$  bonding affects the ability of atoms to rotate about a bond relate molecular polarity to physical properties

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web jul 14 2012 this activity combines two simulations molecular shapes and molecular polarity to guide the students from drawing lewis structures to vsepr predicted geometries to predicting a molecule s polarity subject chemistry level undergrad

intro type homework lab duration 90 minutes answers included no

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web draw lewis structure use vsepr to determine molecular geometry determine bond polarity based on electronegativity differences determine molecular polarity based on bond dipoles molecular geometry for the following molecules complete this

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web molecules composed of covalently bonded atoms may also be polar or nonpolar for the molecule to be polar it must of course have polar bonds but the key factor for determining the polarity of a molecule is its shape if the polar bonds dipoles are symmetrical around the central atom they offset each other and the resulting molecule is

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web linear geometry all angles 180 degrees three regions of electron density around a central atom in a molecule form a trigonal planar geometry all angles 120 degrees four regions of electron density around a central atom in a molecule form a

tetrahedral geometry all angles 109 5 degrees

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web in this investigation you will examine a theory that chemists use to explain different aspects of chemical bonding valence shell electron pair repulsion vsepr theory attention will be given to how molecules are arranged in different shapes and how chemists can predict the geometry of a given molecule