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Chemistry 11

**Classifying Matter**

Assignment

1. Complete the following sentences by filling in the appropriate word from the list below.

Elements	Carbon
Mixtures	Pure
Compound	Letters
Symbols	Carbon dioxide
Properties	

- Elements cannot be separated by physical or chemical means.
- Elements are abbreviated with symbols, which consist of one or two letters.
- Two or more elements chemically combined make up a compound.
- Carbon is an example of an element and carbon dioxide is an example of a compound.
- Elements and compounds are called pure substances because they have a unique set of chemical and physical properties.

2. Fill in the blanks in the following chart.

Element Name	Element Symbol	Common Ions Formed
a. Sodium	Na	+1
b. Copper	Cu	+2, +1
c. Tungsten	W	+6
d. Iron	Fe	+3, +2
e. Tin	Sn	+4, +2
f. Potassium	K	+1
g. Gold	Au	+3, +1
h. Silver	Ag	+1

3. Classify each of the following as one of an Atom (A), a Molecule (M) or an Ion (I).

a. $P^{2-}$	I	c. Ge	A
b. $H_2O$	M	d. O	A
c. $O_2$	M	e. $Ca^{2+}$	I
d. $Au^{2+}$	I	g. $NH_3$	M

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**Herbert Drapkin**

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