

Name: Amanda p7

Properties of Water Worksheet

1. The hydrogen and oxygen atoms are held together by Covalent bonds.
2. The electrons are not shared equally creating a polar molecule.
3. The polarity of water allows it to combine with most substances.
Because of this, it is referred to as the universal solvent.
4. Water molecules stick to other water molecules. This property is called cohesion.
5. Hydrogen bonds form between adjacent water molecules because the positive charged hydrogen end of one water molecule attracts the negative charged oxygen end of another water molecule.
6. Water molecules stick to other materials due to its polar nature. This property is called adhesion.
7. Surface tension creates the skin-like surface formed due to the polar nature of water.
8. Water has a high specific heat, it can absorb a large amount of heat before a change in temperature takes place.
9. Evaporation is when water changes from a liquid to a gas.
10. Condensation is when water changes from a gas to a liquid.
11. Substances that are attracted to water are referred to as hydrophilic and are considered to be water loving.
12. Substances that are not attracted to water are referred to as hydrophobic and are considered to be water fearing.

Word Bank:

polar
cohesion
negative
condensation

hydrophilic
adhesion
high specific heat
hydrophobic

universal solvent
combine

evaporation
covalent
surface tension
positive

13. Draw and label a water molecule.



Identify the positive (+) and negative ends (-).

14. Give the chemical formula for water.



Assessing Water Quality Lab Answer Key

Didier Musso

Assessing Water Quality Lab Answer Key:

Report summaries United States. Environmental Protection Agency,1983 *Selected Water Resources Abstracts* ,1978
Wisconsin Water Quality Assessment Report to Congress Wisconsin. Department of Natural Resources,2002
Inventory of Federal Energy-related Environment and Safety Research for ... ,1980 **Inventory of Federal Energy-related Environment and Safety Research for FY 1979** ,1980 **Oregon's 1992 Water Quality Status Assessment Report** ,1992 Activities of the European Community: General Report of the High Authority European Coal and Steel Community. High Authority,1958 **Environment Information Access** ,1973 Report ,1975 Environment Abstracts Annual 1989 Bowker Editorial Staff,R R Bowker Publishing,Bowker,1990 **Government Reports Announcements & Index** ,1993 **EPA Publications Bibliography** ,2000 *Toxicology Research Projects Directory* ,1980 An indexed directory of current research project abstracts in toxicology and related fields *State of Arizona Water Quality Assessment for 1988, [section 305 (b) Report]* ,1989 **Energy Research Abstracts** ,1990 *The Energy Index* ,1980 **Government Reports Annual Index** ,1994 **Government Reports Announcements** ,1975-05-30 *Report - Hydrodynamics Laboratory, Massachusetts Institute of Technology* Massachusetts Institute of Technology. Hydrodynamics Laboratory,1973-10 **Environmental Assessment of Vancouver Harbour Data Report for the PICES Practical Workshop** Carla M. Stehr,Toshihiro Horiguchi,2001

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In a digital era where connections and knowledge reign supreme, the enchanting power of language has been much more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Assessing Water Quality Lab Answer Key**," compiled by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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