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**Kelly Hackett**

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 $\text{h } 2 \text{ s o } 4 \text{ n a o h n a } 2 \text{ s o } 4 \text{ h } 2 \text{ o}$  if it takes 27 4 ml of 0 768 m naoh n a o h to titrate 16 7 ml of h2so4 h 2 s o 4 what is the concentration of the h2so4 h 2 s o 4 solution hint balance the equation first q2

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 $3 \text{ ag s } 4 \text{ hno } 3 \text{ aq } 3 \text{ agno } 3 \text{ aq } 2 \text{ h } 2 \text{ o l no g a}$

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