



Name: Chelsey Williams Date: 3/31/22

Student Exploration: Stoichiometry

Vocabulary: Avogadro's number, balanced equation, cancel, coefficient, conversion factor, dimensional analysis, formula mass, molar mass, mole, molecular mass, stoichiometry

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. A 250 mL glass of orange juice contains 22 grams of sugar. How much sugar is in a two-liter (2,000 mL) bottle of orange juice? 176 grams
2. It requires two sticks of butter to make a batch of 20 cookies. How much butter will it take to make 150 cookies? 15 sticks of butter

Gizmo Warm-up

Just as a cook follows a recipe to decide how much of each ingredient to add, a chemist uses **stoichiometry** to determine the amounts of substances involved in chemical reactions. The Stoichiometry Gizmo allows you to try your hand at figuring out the amounts of reactants and products that take part in a chemical reaction.

To begin, check that this equation is shown:



1. Look at the **coefficients** (such as the "3" in 3CO) in front of each substance in the equation. The coefficients tell you how many molecules or atoms take part in a chemical reaction. In the spaces below, list the number of each molecule or atom in the equation:

Fe_2O_3 1

CO 3

Fe 2

CO_2 3

2. In a **balanced equation**, the same number of each kind of atom is shown on each side of the equation. Calculate the number of iron (Fe), oxygen (O), and carbon atoms (C).

New equation



New question

How many moles of carbon monoxide (CO) are required to react completely with 1.75 moles of iron (III) oxide (Fe_2O_3)?

Problems solved: 0



Answer Key For Stoichiometry Gizmo

SB Merriam

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the 'SB Merriam' text.

Answer Key For Stoichiometry Gizmo:

Uncover the mysteries within Explore with is enigmatic creation, Discover the Intrigue in **Answer Key For Stoichiometry Gizmo** . This downloadable ebook, shrouded in suspense, is available in a PDF format (*). Dive into a world of uncertainty and anticipation. Download now to unravel the secrets hidden within the pages.

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Table of Contents Answer Key For Stoichiometry Gizmo

1. Understanding the eBook Answer Key For Stoichiometry Gizmo
 - The Rise of Digital Reading Answer Key For Stoichiometry Gizmo
 - Advantages of eBooks Over Traditional Books
2. Identifying Answer Key For Stoichiometry Gizmo
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answer Key For Stoichiometry Gizmo
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answer Key For Stoichiometry Gizmo
 - Personalized Recommendations
 - Answer Key For Stoichiometry Gizmo User Reviews and Ratings
 - Answer Key For Stoichiometry Gizmo and Bestseller Lists
5. Accessing Answer Key For Stoichiometry Gizmo Free and Paid eBooks
 - Answer Key For Stoichiometry Gizmo Public Domain eBooks
 - Answer Key For Stoichiometry Gizmo eBook Subscription Services
 - Answer Key For Stoichiometry Gizmo Budget-Friendly Options

6. Navigating Answer Key For Stoichiometry Gizmo eBook Formats
 - ePub, PDF, MOBI, and More
 - Answer Key For Stoichiometry Gizmo Compatibility with Devices
 - Answer Key For Stoichiometry Gizmo Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answer Key For Stoichiometry Gizmo
 - Highlighting and Note-Taking Answer Key For Stoichiometry Gizmo
 - Interactive Elements Answer Key For Stoichiometry Gizmo
8. Staying Engaged with Answer Key For Stoichiometry Gizmo
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answer Key For Stoichiometry Gizmo
9. Balancing eBooks and Physical Books Answer Key For Stoichiometry Gizmo
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answer Key For Stoichiometry Gizmo
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Answer Key For Stoichiometry Gizmo
 - Setting Reading Goals Answer Key For Stoichiometry Gizmo
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Answer Key For Stoichiometry Gizmo
 - Fact-Checking eBook Content of Answer Key For Stoichiometry Gizmo
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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