

Name : \_\_\_\_\_

Date : \_\_\_\_\_

## STOICHIOMETRY PRACTICE WORKSHEET

1. Using the following equation:



Calculate how many grams of iodine are needed to prepare 28.6 grams of ICl by this reaction.

2. Using the following equation:



How many moles and grams of KMnO<sub>4</sub> are needed for this reaction on 11.4 grams of KNO<sub>2</sub>?

3. Using the following equation:



How many moles and grams of oxygen (O<sub>2</sub>) are needed to react with 56.8 grams of ammonia by this reaction?

4. Using the following equation:



Calculate the number of moles and the number of grams of iodine (I<sub>2</sub>) that can be made this way from 16.4 grams of NaIO<sub>3</sub>.

# 9 Stoichiometry Practice Problems Review Answers

**IM Harris**

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