
Reinforce and Review

Lesson Worksheets

Copy and distribute the Lesson 12.3 worksheets in CK-12 Chemistry –Intermediate Workbook. Ask students to complete the worksheets to reinforce lesson content.

Lesson Review Questions

Have students answer the Lesson Review Questions at the end of Lesson 12.3 in CK-12 Chemistry –Intermediate FlexBook® resource.

Sample Answers

Practice Problems

1.

$$1.30 \text{ mol } C_3H_8 \times \frac{5 \text{ mol } O_2}{1 \text{ mol } C_3H_8} = 6.50 \text{ mol } O_2 \text{ needed}$$

Since the moles of O_2 needed to fully react with all of the C_3H_8 is more than the amount of O_2 present (6.00 mol), O_2 is the limiting reactant.

2.

$$8.00 \text{ g } Zn \times \frac{1 \text{ mol } Zn}{65.38 \text{ g } Zn} \times \frac{2 \text{ mol } HCl}{1 \text{ mol } Zn} \times \frac{36.46 \text{ g } HCl}{1 \text{ mol } HCl} = 8.92 \text{ g } HCl \text{ needed}$$

Since there is less HCl present (8.00 g) than what is needed to react with all of the Zn, the HCl is the limiting reactant. This problem can also be solved by calculating the moles of Zn and the moles of HCl originally present. Compare the number of moles, taking note of the 1:2 ratio.

$$8.00 \text{ g } Zn \times \frac{1 \text{ mol } Zn}{65.38 \text{ g } Zn} = 0.122 \text{ mol } Zn$$
$$8.00 \text{ g } HCl \times \frac{1 \text{ mol } Zn}{36.46 \text{ g } HCl} = 0.219 \text{ mol } HCl$$

The moles of HCl is not double the moles of Zn, as required by the balanced equation. Therefore, HCl is the limiting reactant.

3.

- The balanced equation requires twice as many moles of Na as Cl_2 . Sodium (Na) is the limiting reactant.
- The 3.40 moles of Na can be converted to moles of Cl_2 that would react.

$$3.40 \text{ mol } Na \times \frac{1 \text{ mol } Cl_2}{2 \text{ mol } Na} = 1.70 \text{ mol } Cl_2 \text{ reacts}$$

The excess Cl_2 is determined by subtraction. $4.20 \text{ mol} - 1.70 \text{ mol} = 2.50 \text{ mol}$ excess Cl_2

- $3.40 \text{ mol } Na \times \frac{2 \text{ mol } NaCl}{1 \text{ mol } Na} = 3.40 \text{ mol } NaCl$

4. First, the limiting reactant must be determined.

$$35.0 \text{ g } N_2 \times \frac{1 \text{ mol } N_2}{28.02 \text{ g } N_2} \times \frac{3 \text{ mol } H_2}{1 \text{ mol } N_2} \times \frac{2.02 \text{ g } H_2}{1 \text{ mol } H_2} = 7.57 \text{ g } H_2 \text{ needed}$$

12 3 Review And Reinforce Answers

Cynthia Y. Young



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the mark of vishnu summary studienet dk - Jan 28 2022

web summary the short story the mark of vishnu by khushwant singh begins with gunga ram a servant leaving a saucer of milk for a cobra he calls the kala nag outside a hole in the wall he tells the schoolboys the milk is gone by morning and this offering protects the household from the snake

the mark of vishnu characters studienet dk - Dec 27 2021

web the main characters in the short story the mark of vishnu by khushwant singh are gunga ram and the four schoolboys
gunga ram is a servant in the household where the schoolboys live the four brothers go to school and receive a formal education

the mark of vishnu by khushwant singh the sitting bee - May 12 2023

web in the mark of vishnu by khushwant singh we have the theme of tradition modernity faith loyalty respect and change
taken from his collected stories collection the story is narrated in the first person by a young unnamed boy and after reading the story the reader realises that singh may be exploring the theme of tradition

the mark of vishnu themes and message prime study guides - Apr 30 2022

web the theme of the short story the mark of vishnu by khushwant singh is superstition and ignorance the main character
gunga ram is both superstitious and ignorant he believes that by leaving a saucer full of milk for the snake to drink he is
protecting the household from snake bites

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web the main characters in the short story the mark of vishnu by khushwant singh are gunga ram and the four schoolboys
gunga ram is a servant in the household where the schoolboys live the four brothers go to school and receive a formal
education since gunga ram is a servant the schoolboys make fun of him for his lack of education and