

- According to the Bronsted—Lowry definition, which chemical species can function *both* as an acid and as a base?
 - Cl^-
 - SO_4^{2-}
 - NH_4^+
 - HCO_3^-
 - H_3O^+
- In the reaction $\text{CN}^- + \text{H}_2\text{O} \rightleftharpoons \text{HCN} + \text{OH}^-$ which is an acid—base conjugate pair?
 - H_2O and HCN
 - H_2O and OH^-
 - CN^- and H_2O
 - HCN and OH^-
- Given that HX is a stronger Bronsted acid than HY in aqueous solution, which is true of a 1 M solution of NaX ?
 - It is less basic than a 1 M solution of NaY .
 - It is more basic than a 1 M solution of NaY .
 - It yields a neutral solution.
 - It is more concentrated than a 1 M solution of NaY .
- The pH of a 0.03 M HCl solution is
 - 1.5
 - 2.5
 - 3.5
 - 12.5
- The pH of a 1.0×10^{-2} M $\text{Ba}(\text{OH})_2$ solution at 25°C is
 - 2.7
 - 3.0
 - 11.0
 - 11.3
- Which is the strongest acid?
 - HClO
 - HClO_2
 - HClO_3
 - HClO_4
- Which salt reacts with water (hydrolyzes) to produce a basic solution?
 - $\text{NaC}_2\text{H}_3\text{O}_2$
 - NaNO_3
 - NH_4Cl
 - BaSO_4

- The *weakest* of the bases listed is

Acid	Conjugate Base	K_a (Ionization Constant of Acid)
HCl	Cl^-	100% ionized
HSO_4^-	SO_4^{2-}	1.2×10^{-2}
H_2S	HS^-	5.7×10^{-8}
HS^-	S^{2-}	1.2×10^{-13}

 - Cl^-
 - CN^-
 - HS^-
 - S^{2-}
 - SO_4^{2-}

- The oxide of which element will react with water to form the strongest acid?

	Main Groups							
Period	I	II	III	IV	V	VI	VII	O
First								
Second	X	Y	Z	P		Q	S	U
Third	W					R	T	M

 - W
 - M
 - P
 - R
 - Z

- Which statement is a logical inference from the fact that a 0.10 M solution of potassium acetate $\text{KC}_2\text{H}_3\text{O}_2$ is less alkaline than a 0.10 M solution of potassium cyanide, KCN ?
 - Hydrocyanic acid is a weaker acid than acetic acid.
 - Hydrocyanic acid is less soluble in water than acetic acid.
 - Cyanides are less soluble than acetates.
 - Acetic acid is a weaker acid than hydrocyanic acid.
- In the titration of 50.0 mL of 0.100 M benzoic acid (a monoprotic acid) with 50.0 mL of 0.100 M NaOH , the properties of the solution at the equivalence point will correspond exactly to the properties of
 - a 0.100 M sodium solution.
 - a 0.0500 M sodium hydroxide solution.
 - a 0.0500 M benzoic acid solution.
 - a 0.0500 M sodium benzoate solution.

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Sustainable Development Research Walter Leal Filho,Amanda Lange Salvia,Claudio Ruy Portela de

Vasconcelos,2024-09-20 This book involves establishing a set of priorities and a roadmap that can guide scholarly and practical efforts towards sustainability goals It encourages collaboration across disciplines to address complex sustainability issues that span social economic and environmental domains It also supports the development of robust methodologies for conducting research including quantitative qualitative and mixed methods approaches But despite the importance of and the need for an agenda for sustainable development research many efforts are isolated and thematically disconnected Also it is difficult to find information on how sustainability research is being undertaken and on the wide range of methods being used

Against this backdrop and in order to facilitate a broad discussion on the contribution of sustainable development research this book is being produced The book gathers inputs from universities and research organisations working on matters related to sustainable development research in a variety of contexts It also provides a platform for the dissemination of information on the latest initiatives paving the way for technology transfer and networking Furthermore the book intends to provide a fertile basis upon which universities research centres and practitioners may cooperate more closely in this key area Last but not least a further aim of the book is to present methodological approaches and experiences deriving from case studies and projects which aim to show how sustainability research may be implementing across a range of disciplines Thanks to its scope and interdisciplinarity this books makes an excellent reading to everyone interested on sustainability research

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