

# ACIDS, BASES AND SALTS

<b>Definition of an acid</b> An acid is a compound which when dissolved in water produces hydrogen ions as the only positively charged ions	<b>What causes the acidic properties of acids?</b> The hydrogen ions ( $H^+$ ) cause the acidic properties and these are formed in the presence of water.
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<b>Another term to refer to an acid</b> An acid is called a proton donor	<b>Why an acid is also called a proton donor?</b> It's because an acid provides protons or hydrogen ions ( $H^+$ ) to other substances during the reaction.
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<b>Substance to which an acid provides protons</b> BASE	<b>A base is then called</b> PROTON ACCEPTOR	<b>Why the base is called a proton acceptor?</b> Because the base accepts hydrogen ions from acids	<b>Equation for the reaction between acid and base</b> $H_{(aq)} + OH_{(aq)} \longrightarrow H_2O_{(l)}$
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<b>Common laboratory acids</b>		<b>These three common laboratory acids are also called</b> MINERAL ACIDS	<b>Why these acids are also called mineral acids?</b> They are derived from mineral salts ie chlorides for $HCl$ , sulphates for $H_2SO_4$ and nitrates for $HNO_3$
Hydrochloric acid	$HCl$		
Sulphuric acid	$H_2SO_4$		
Nitric acid	$HNO_3$		

<b>Other mineral acids known</b>	<b>Mineral salts from which the acid is derived</b>	<b>Organic acids known</b>	<b>Naturally occurring acids</b>
Sulphurous acid	$H_2SO_3$	Derived from sulphites	CITRIC ACID from lemons
Carbonic acid	$H_2CO_3$	Derived from carbonates	TARTARIC ACID from grapes
Phosphoric acid	$H_3PO_4$	Derived from phosphates	ACETIC ACID from vinegar
Nitrous acid	$HNO_2$	Derived from nitrates	LACTIC ACID from sour milk
		Methanoic acid ( $HCOOH$ )	Hydrochloric acid from digestive juices

<b>Whenever an acid is dissolved in water, it produces</b> HYDROGEN IONS	<b>Term given to the number of hydrogen ions produced by one molecule of an acid</b> BASICITY OF AN ACID	<b>Definition of basicity of an acid</b> BASICITY of an acid is the <u>number of hydrogen ions produced by one molecule of an acid in aqueous solution</u> .
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# Acids And Bases Study Guide

**Thomas Elliott Taylor**

## Acids And Bases Study Guide:

*Organic Chemistry, 12e Binder Ready Version Study Guide & Student Solutions Manual* T. W. Graham Solomons, Craig B. Fryhle, Scott A. Snyder, 2016-04-11 This is the Student Study Guide Solutions Manual to accompany Organic Chemistry 12th Edition The 12th edition of Organic Chemistry continues Solomons Fryhle Snyder's tradition of excellence in teaching and preparing students for success in the organic classroom and beyond A central theme of the authors approach to organic chemistry is to emphasize the relationship between structure and reactivity To accomplish this the content is organized in a way that combines the most useful features of a functional group approach with one largely based on reaction mechanisms The authors philosophy is to emphasize mechanisms and their common aspects as often as possible and at the same time use the unifying features of functional groups as the basis for most chapters The structural aspects of the authors approach show students what organic chemistry is Mechanistic aspects of their approach show students how it works And wherever an opportunity arises the authors show students what it does in living systems and the physical world around us *Organic Chemistry, 13e Student Study Guide and Solutions Manual* T. W. Graham Solomons, Craig B. Fryhle, Scott A.

Snyder, 2022-05-03 Organic Chemistry Student Study Guide and Solutions Manual 13th Edition offers the full solutions for select exercises from the text **Study Guide for Chemical Principles** Thomas Elliott Taylor, 1979 Acid-Base Equilibria - Quick Chemistry Review Outline and Handout E Staff, Acid Base Equilibria Quick Review Outline and Handout for All Students Learn and review on the go Use Quick Review Chemistry Notes to help you learn or brush up on the subject quickly You can use the review notes as a reference to understand the subject better and improve your grades Easy to remember facts to help you perform better Perfect study notes for all high school and college students 10 Pages Basic Concepts of Chemistry, Study Guide Leo J. Malone, 2000 The text's three main goals are to introduce chemistry as a living relevant science to encourage learning and critical thinking and to help readers overcome the math difficulties that impede their progress in chemistry Designed to help readers master the principles of general chemistry As a prep book it promotes active involvement with the material There are special features throughout that reinforce concepts and help to develop strong problem solving and study skills Updated to Include an Interactive Learning Ware problems CD containing several of the chapter ending problems from the book in an interactive tutorial with feedback to help readers set up and solve problems

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*Qualitative Chemical Analysis. A Guide in the Practical Study of Chemistry and in the Work of Analysis ... Second Edition, Revised* Silas H. DOUGLAS (and PRESCOTT (Albert B.)),1876 *Organic Chemistry, Study Guide and Solutions Manual* T. W. Graham Solomons,Craig B. Fryhle,2000 On the cover of this book is a Pacific yew tree found in the ancient forests of the Pacific Northwest The bark of the Pacific yew tree produces Taxol found to be a highly effective drug against ovarian and breast cancer Taxol blocks mitosis during eukaryotic cell division The supply of Taxol from the Pacific yew tree is vanishingly small however A single 100 year old tree provides only about one dose of the drug roughly 300 mg For this reason as well as the spectacular molecular architecture of Taxol synthetic organic chemists fiercely undertook efforts to synthesize it Five total syntheses of Taxol have thus far been reported Now a combination of isolation of a related metabolite from European yew needles and synthesis of Taxol from that intermediate supply the clinical demand This case clearly demonstrates the importance of synthesis and the use of organic chemistry It s just one of the many examples used in the text that will spark the interest of students and get them involved in the study of organic chemistry **Study Guide to Accompany Introduction to General, Organic, & Biochemistry** Frederick A. Bettelheim,William M. Scovell,1998 **Student Study Guide and Solutions Manual to Accompany General, Organic, and Biochemistry** Katherine J. Denniston,Denniston Katherine,2006

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