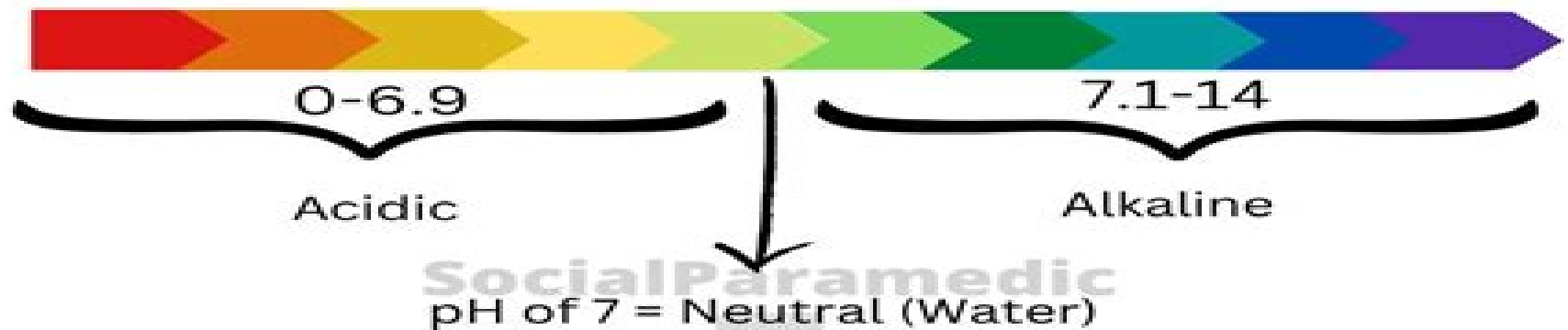


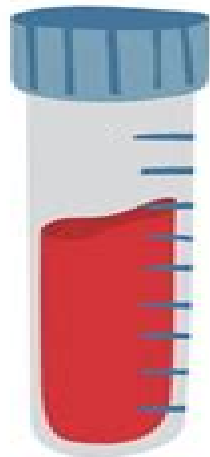
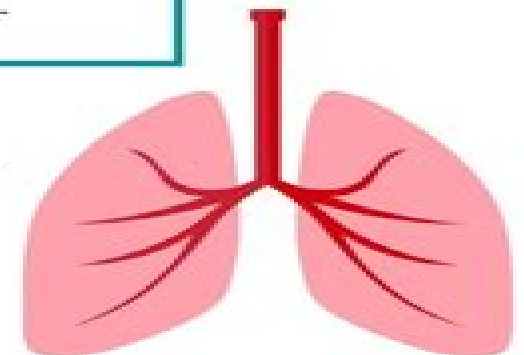
# Acid-Base Balance



## Normal Ranges

pH of Blood	7.35-7.45
ETCO <sub>2</sub>	35-45mmHg
HCO <sub>3</sub> (Bicarbonate)	22-26 meq/L

**HCO<sub>3</sub>**: Normally, O<sub>2</sub> combines with glucose to create energy and CO<sub>2</sub> is an acid waste by product of this. Bicarbonate is a buffer produced by the body that is alkaline, and binds with CO<sub>2</sub> to neutralize it to allow the body to get rid of it.



# Acid Base Study Guide Answers

**Barbara Janson Cohen, Memmler**



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