

ACS EXAM (2024 version) Practice Questions and Answers

A 20.0 g sample of mercury(II) oxide (HgO , $M = 216.6$)

is heated strongly, causing it to decompose to metallic Hg and O_2 gas. What volume of O_2 gas is produced (measured at STP)?

- (A) 1.03 L
- (B) 2.07 L
- (C) 4.14 L
- (D) 14.0 L - ✓✓Correct-A

When 30.0 mL of 0.10 M AgNO_3 is added to 30.0 mL of 0.10 M NaCl , aqueous NaNO_3 and solid AgCl are formed. How much solid AgCl is produced?

- (A) 0.0030 mol
- (B) 0.0060 mol
- (C) 0.030 mol
- (D) 0.060 mol - ✓✓Correct-A

How much $\text{Sr}(\text{OH})_2 \cdot 8 \text{H}_2\text{O}$ ($M = 265.76$) is needed to

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S Ashworth



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