

ACS General Chemistry Final Formula Sheet

1. Atomic Structure & Quantum Mechanics

- Average Atomic Mass = Σ (mass of isotope $\times$ relative abundance)
- Speed of Light Equation: $c = \lambda \nu$
- Photon Energy: $E = h\nu$

2. Chemical Bonding & Molecular Properties

- Formal Charge = (valence electrons) - (non-bonding electrons) - $\frac{1}{2}$ (bonding electrons)

3. Solutions

- Molarity (M) = moles of solute / liters of solution
- Molality (m) = moles of solute / kilograms of solvent
- Mole Fraction (X) = moles of A / total moles
- Freezing Point Depression: $\Delta T_f = i K_f m$
- Boiling Point Elevation: $\Delta T_b = i K_b m$
- Dilution Formula: $M_1 V_1 = M_2 V_2$

4. Stoichiometry

- Percent Yield = (actual yield / theoretical yield) $\times$ 100%

5. Thermochemistry (Energetics)

- Heat Transfer: $q = mc\Delta T$
- Enthalpy of Reaction from ΔH°_f : $\Delta H^\circ_{\text{rxn}} = \Sigma \Delta H^\circ_f(\text{products}) - \Sigma \Delta H^\circ_f(\text{reactants})$
- Enthalpy of Reaction from Bond Energies: $\Delta H^\circ_{\text{rxn}} = \Sigma \text{Bond Energies}(\text{reactants}) - \Sigma \text{Bond Energies}(\text{products})$
- Gibbs Free Energy: $\Delta G = \Delta H - T\Delta S$
- Relationship between ΔG° and K: $\Delta G^\circ = -RT \ln K$

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Sebastian Brünink

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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web untuk bilangan berpangkat positif itu sendiri memiliki beberapa sifat tertentu dimana bilangannya terdiri atas a b sebagai bilangan real dan m n adalah bilangan bulat positif adapun sifat sifat khusus yang dimiliki oleh bilangan berpangkat positif adalah sebagai berikut $a^m \times a^n = a^{m+n}$ $a^m \div a^n = a^{m-n}$ berlaku untuk m n serta $b \neq 0$

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