

ACS ANALYTICAL CHEMISTRY STUDY GUIDE

- k_f' : $k_f' = k_f^* \cdot a$
- EDTA titrations: 1. before eq EDTA binds to metal 1:1
- 2. At eq we assume that all metal ions can bind to EDTA and then use k_f'
- Potentiometry: an analytical method in which an electric potential difference (avoltage) of a cell is measured

- Reducing agent: Donates electrons and becomes oxidized.
- oxidizing agent: Accepts electrons and becomes reduced.
- coulometry: science of measuring change and relating it to analyte
- cathode: the electrode at which reduction occurs
- Anode: the electrode at which oxidation occurs
- Ecell: $E_{cell} = E_{red} - E_{ox}$
- Faraday's Equation: $E = E - (0.0592/n) \log(a/b)$
- ||: anode always written on left of ||
- amperometry: elec techniques where the cell current is measured where pot diff between the indicator and reference electrode is controlled
- Clark electrode: Electrode for oxygen
- Biosensors: Detecting devices that can detect specific substances in cells
- amp chromo det: point, size, or charge
- limiting current: the diff between current and steady state max
- Anodic vol: Very sensative for analyzing metals into ppb
- fluroescence: used to measure singlets, atoms like rigid aromatic rings, high sensitivity
- phosphorescence: involves emission of light trip- singlet, much lower than fluro
- Transmittance: $A = -\log(T)$ or $2 - \log(T\%)$
- atomic absorbtion: lamp-flame-mono-detector
- atomic emission: flame-mono-detector
- A.A. hollow cathode lamp: used to form very narrow band lights
- A.A. atomizer: Heat sample to very high temps to break all chemical bonds, hot enough to

Acs Chemistry Study Guide

**K. Peter C. Vollhardt, Neil E.
Schore, American Chemical Society**

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and vocabulary that the American Chemical Society ACS Examinations Institute expects you to have mastered before sitting for your exam Test sections include Atoms Properties of Matter Bonding and Intermolecular Interactions Reactions Kinetics and Equilibrium Acids and Bases Thermodynamics Electrochemistry Nuclear Chemistry Safety Math and Data in the Laboratory and much more Our guide is full of specific and detailed information that will be key to passing your exam Concepts and principles aren't simply named or described in passing but are explained in detail The Mometrix general chemistry study guide is laid out in a logical and organized fashion so that one section naturally flows from the one preceding it Because it's written with an eye for both technical accuracy and accessibility you will not have to worry about getting lost in dense academic language Any test prep guide is only as good as its practice questions and answer explanations and that's another area where our guide stands out The Mometrix test prep team has provided plenty of general chemistry practice test questions to prepare you for what to expect on the actual exam Each answer is explained in depth in order to make the principles and reasoning behind it crystal clear Many concepts include links to online review videos where you can watch our instructors break down the topics so the material can be quickly grasped Examples are worked step by step so you see exactly what to do We've helped hundreds of thousands of people pass standardized tests and achieve their education and career goals We've done this by setting high standards for Mometrix Test Preparation guides and our ACS General Chemistry Study Guide ACS Exam Prep Secrets is no exception It's an excellent investment in your future Get the general chemistry review you need to be successful on your exam

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