

Name \_\_\_\_\_

### Acids and bases end of unit worksheet

1. Complete the table below. I can identify properties of acids and bases.

	Acids	Bases
Taste	Sour	bitter
pH range	0-7	7-14
Color with litmus paper	pink/red	blue
Are they electrolytes? (yes or no)	yes	yes
What do they react with?	carbonates metals bases	fats+oils acids
What ion do they produce?	$H^+$ or $H_3O^+$	$OH^-$
How do they feel?	corrosive (itch, burn)	Slippery
Color with phenolphthalein	colorless	pink

2. I can list uses of acids in industry. List two uses of acids in industry.

Cleaning concrete (HCl) Making paint, plastics, fertilizers  
producing batteries

3. I can list examples of common acids and bases found in a home or store.

List 2 common acids and 2 common bases you might find in your home or at a store.

Acids: lemons, vinegar, sour candy, pop, aspirin, orange  
juice, lemonade

Bases: drain cleaner, soap, baking soda, ammonia,  
window cleaner, tums, antacids

4. I can name acids and bases. Name the following:

HF hydrofluoric Acid  $HNO_2$  Nitrous acid  $Ca(OH)_2$  calcium hydroxide

Write the formula for the following: (Check your charges!)

Magnesium hydroxide  $Mg(OH)_2$  Hydrobromic acid HBr carbonic acid  $H_2CO_3$

5. I can define and identify Arrhenius acids and bases. Label the following as an acid (A), base (B) or neutral (N).

$H_3PO_4$  A  $HCl$  A  $BaBr_2$  N  $LiOH$  B

6. I can label an acid as mono, di, or tri protic. Label the following as mono, di, or triprotic.

$H_2SO_4$  diprotic  $H_3PO_4$  triprotic  $HNO_3$  monoprotic  $HBr$  monoprotic

# Acids Bases And Solutions Key Terms Answers

**Steven S. Zumdahl**



## **Acids Bases And Solutions Key Terms Answers:**

*Fundamentals of Nursing - E-Book* Barbara L Yoost, Lynne R Crawford, 2015-02-24 There is a new fundamentals text in town One that centers on simple language active learning and a fresh new way to help you truly understand apply and retain important nursing information and concepts Introducing the brand new Fundamentals of Nursing text from Yoost and Crawford Written in a warm and conversational style this innovative text starts by guiding you towards a basic understanding of the nursing profession and then logically progresses through the nursing process and into the safe and systematic methods of applying care Each chapter features realistic case studies and critical thinking exercises woven throughout the content to help you continually apply what you've learned to actual patient care Conceptual care maps further your ability to make clinical judgments and synthesize knowledge as you develop plans of care after analyzing and clustering related patient assessment data All of this paired with a wealth of student friendly learning features and clinically focused content offers up a fundamentally different and quite effective way for you to easily master the fundamentals of nursing

*Foundations of College Chemistry* Morris Hein, Susan Arena, Cary Willard, 2016-08-02 This text is an unbound three hole punched version Used by over 750 000 students Foundations of College Chemistry Binder Ready Version 15th Edition is praised for its accuracy clear no nonsense approach and direct writing style Foundations direct and straightforward explanations focus on problem solving making it the most dependable text on the market Its comprehensive scope proven track record outstanding in text examples and problem sets were all designed to provide instructors with a solid text while not overwhelming students in a difficult course Foundations fits into the prep intro chemistry courses which often include a wide mix of students from science majors not yet ready for general chemistry allied health students in their 1st semester of a GOB sequence science education students for elementary school teachers to the occasional liberal arts student fulfilling a science requirement Foundations was specifically designed to meet this wide array of needs

*Principles of Anatomy and Physiology, 4th Asia-Pacific Edition* Gerard J. Tortora, Bryan H. Derrickson, Brendan Burkett, Julie Cooke, Flavia DiPietro, Tara Diversi, Danielle Dye, Alexander Engel, Hayley Green, Michael Macartney, Mark McKean, Gregory Peoples, Simon Summers, 2025-10-10

**Understanding Chemistry** Fred M. Dewey, 1994 *Study Guide with Selected Solutions* Karen Timberlake, 2002-08-06 Written by the author the Study Guide is keyed to the learning goals in the text and designed to promote active learning through a variety of exercises with answers and mastery exams Also contains complete solutions to odd numbered problems

*Basic Concepts of Chemistry* Leo J. Malone, Theodore O. Dolter, 2011-12-27 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end each section and relevant chapter problems at the end of each chapter Every concept in the text is clearly illustrated with one or more step by step examples Making it Real

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**General Chemistry** Ralph H. Petrucci, William S. Harwood, 1993 General Chemistry Principles and Modern Applications is recognized for its superior problems lucid writing and precision of argument This updated and expanded edition retains the popular and innovative features of previous editions including Feature Problems follow up Integrative and Practice Exercises to accompany every in chapter Example and Focus On application boxes as well as new Keep in Mind marginal notes Topics covered include atoms and the atomic theory chemical compounds and reactions gases Thermochemistry electrons in atoms chemical bonding liquids solids and intermolecular forces chemical kinetics principles of chemical equilibrium acids and bases electrochemistry representative and transitional elements and nuclear and organic chemistry For individuals interested in a broad overview of chemical principles and applications

*Student Study Guide and Solutions Manual to Accompany General, Organic, and Biochemistry* Katherine J. Denniston, Denniston Katherine, 2006

*Chemical Interactions*, 2005

**Introduction to Modern Chemistry: Student Manual** Eugene Meyer, 1979

*Basic Chemistry* Steven S. Zumdahl, 1999 For a full description see catalog entry for Zumdahl Introductory Chemistry A Foundation 4 e

**Chapter Resource 2 Chemistry of Life Biology** Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2004

**Introduction to General, Organic & Biochemistry** Frederick A. Bettelheim, Jerry March, 1991

**Fundamentals of Biochemistry** Destin Heilman, Stephen Woski, Donald Voet, Judith G. Voet, Charlotte W. Pratt, 2024-03-28 Fundamentals of Biochemistry 6th edition with new authors Destin Heilman and Stephen Woski provides a solid biochemical foundation that is rooted in chemistry while presenting complete and balanced coverage that is clearly written and relevant to human health and disease This edition includes new pedagogy and enhanced visuals that better adapt the text for the modern student including a focus on enhanced self assessment tools and scaffolding of learning outcomes throughout the text The new authors continue the trusted pedagogy of the previous five editions and present approachable balanced coverage to provide students with a solid biochemical foundation to prepare them for future scientific challenges The pedagogy remains focused on biochemistry s key theme the relationship between structure and function while streamlining the student experience to better focus attention on the critical subject matter Fundamentals of Biochemistry 6e includes a significant update to the art program with modernized more effective renderings that better enable understanding of the subject matter New scaffolded learning outcomes in each section and a focus on self assessment tools both streamline and elevate the effectiveness of the new edition as a critical learning resource for biochemistry students

**Elements of General and Biological Chemistry** John R. Holum, 1987 This book is a shorter version of the third edition of Fundamentals of General Organic and Biological Chemistry 1986 It incorporates the recommendations of the Task Force on Chemical Education for Health Professions and meets the needs for

a basic text in a one term course in chemistry for students aiming for careers in professional health care fields

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