

Acids and Bases I: Webquest

On Your Desk:

- ❖ This packet
- ❖ Pencil
- ❖ Calculator

Expectations:

- ❖ Students will work quietly and respectfully.
- ❖ Students will raise their hands before speaking
- ❖ Students will not leave their seats without permission
- ❖ Teacher will respect and encourage all questions
- ❖ Teacher will enforce all rules and expectations

Introduction

Acids and bases are everywhere!

You use chemicals all the time! The deodorant, shampoo, face wash, etc. that you use are all made of chemicals, and they act differently from one another. You wouldn't use vinegar to wash your hair, or laundry detergent in the dishwasher. Would you use toothpaste to flavor a cake? Would you use toilet bowl cleaner to perm your hair? Would you use lemonade as a replacement for chicken broth? When you have heartburn, would you take TUMS or drink orange juice?

Come up with your own Ridiculous Substitution here! _____

WHY, though, are all these household compounds so different from one another? Can we learn enough about their chemistry to understand why some household materials act the way that they do?

In today's activity, we will learn that various household chemicals have different pHs.

What is pH?

The power of an acid or base is measured with the pH scale.

pH means "the power of hydrogen". The pH tells you how many hydrogen ions are in a solution. This affects a solution in many ways – from how it feels and tastes to how it will react chemically.

Take a moment and remember what kind of ion hydrogen makes. Look at the periodic table if you need to!

What ion does hydrogen form? _____

The pH scale goes from 1-14. Water is neutral, and has a pH of 7. Acids have a pH below 7, and bases have a pH above 7. Each value of pH means the concentration of H^+ changes by a power of 10. This means that a solution with a pH of 6 has 10 times more H^+ ions in it than a solution with a pH of 7. As the concentration of H^+ ions increases, the concentration of OH^- ions decreases, and vice versa. At pH 7, the concentrations of H^+ and OH^- are equal.



These acids belong to the 20 student acids program.

Acids And Bases Webquest Answers

Lei Huang

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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