

# Acids & Bases: Assessment Review

Name \_\_\_\_\_

1. **True/False** – Write True if the statement is true, False if the statement is false, Conjugate Acid and Conjugate Base if an acid and a base are mixed together.

2. Rewrite a chemical equation showing how  $\text{HClO}_4$  behaves as an acid when dissolved in water. Label the acid, base, conjugate acid & conjugate base.

3. Rewrite a chemical equation showing how ammonia ( $\text{NH}_3$ ) behaves as a base when dissolved in water. Label the acid, base, conjugate acid & conjugate base.

4. **True/False** – Write True if the calculation with either  $[\text{H}^+]$  or  $[\text{OH}^-]$  shows the concentration based on pH from pOH (or vice versa pH from pOH calculation.)

5. Recreate the calculation flowchart on the diagram provided. (be careful with the placement!)



6. Calculate  $[\text{OH}^-]$  in a solution that  $[\text{H}^+] = 7.2 \times 10^{-7}$  M. Is the solution acidic, basic, or neutral?

7. Calculate  $[\text{H}^+]$  in a solution of  $[\text{OH}^-] = 5.2 \times 10^{-11}$  M. Is the solution acidic, basic, or neutral?

8. Calculate the  $[\text{H}^+]$  in a solution that shows a pH of 1.1. Is the solution acidic, basic, or neutral?

9. Calculate the  $[\text{OH}^-]$  in a solution of  $[\text{OH}^-] = 1.2 \times 10^{-9}$  M if the solution acidic, basic, or neutral?

# Acids And Bases Review Answer Key Chemistry

**Fred M. Dewey**

## Acids And Bases Review Answer Key Chemistry:

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