

Acids and Bases Problems

1. Rank the following solutions in order of increasing acidity (from lowest to highest):

Solution A with pH = 6.50

Solution C with $[\text{OH}^-] = 6.0 \times 10^{-10} \text{ M}$

Solution B with $[\text{H}_3\text{O}^+] = 3.5 \times 10^{-3} \text{ M}$

Solution D with pH = 5.85

_____ < _____ < _____ < _____

2. Complete the following table.

Acid	Conjugate Base
HS^-	
	ClO_4^-
H_3PO_4	
	HClO_3^-

3. Write the acid-base reactions with water for each of the following acids.

Label the acid, base, conjugate acid, and conjugate base.



Acid Bases And Salts Practice Problems Answers

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Slowinski, Conrad L. Stanitski**

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