



pH Worksheet



1. Consider the dissociation of aluminum hydroxide. If the pH is 9.85, what is the concentration of the aluminum hydroxide solution?
2. Consider the dissociation of calcium hydroxide. If the pH is 11.64 and you have 2.55 L of solution, how many grams of calcium hydroxide are in the solution?
3. What is the pH of a solution with a $[\text{OH}^-]$ of 5.7×10^{-5} M?
4. A solution has a pH of 6.1. What is the concentration of hydroxide ions?
5. A solution is created by measuring 3.60×10^{-3} moles of NaOH and 5.95×10^{-4} moles of HCl into a container. Then, water is added until the final volume is 1.00 L. What is the pH of this solution?
6. A solution with a H^+ concentration of 1.00×10^{-7} M is considered neutral. Why?

Acid Base Water Ph And Poh Worksheet Reteach Answers

Annelies Wilder-Smith

Acid Base Water Ph And Poh Worksheet Reteach Answers:

Uncover the mysteries within is enigmatic creation, Discover the Intrigue in **Acid Base Water Ph And Poh Worksheet Reteach Answers**. This downloadable ebook, shrouded in suspense, is available in a PDF format (PDF Size: *). Dive into a world of uncertainty and anticipation. Download now to unravel the secrets hidden within the pages.

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