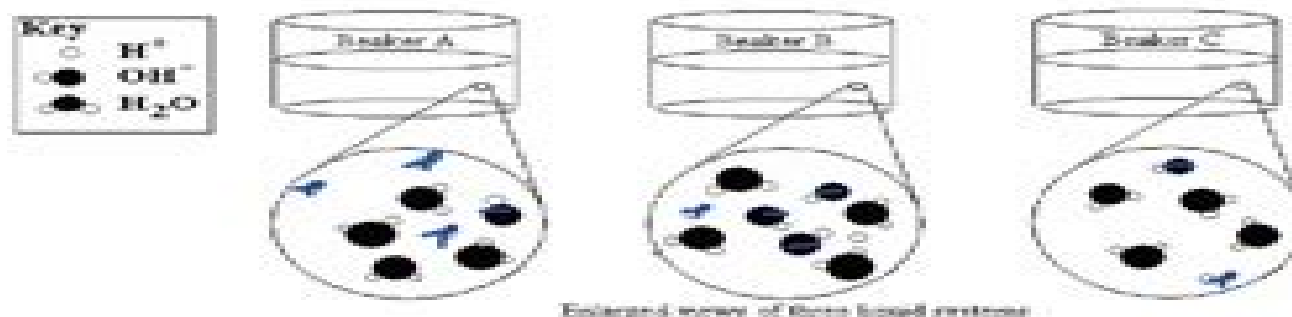


## Acid-Base Neutralization Reactions

**Model:**

Note: spectator ions are not shown in this model, but they are present in each solution.

**Key Questions**

1. How does the concentration of  $\text{H}^+$  compare to the concentration of  $\text{OH}^-$  in solution A?

**3:1**

2. How does the concentration of  $\text{H}^+$  compare to the concentration of  $\text{OH}^-$  in solution B?

**1:3**

3. How does the concentration of  $\text{H}^+$  compare to the concentration of  $\text{OH}^-$  in solution C?

**1:1**

4. Identify the acidic solution in the model.

**A**

5. Identify the basic solution in the model.

**B**

6. Identify the neutral solution in the model.

**C**

# Acid Base Neutralization Pogil Answers

**Arshad Iqbal**



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