

## Chemistry CST Released Questions

### Multiple Choice

Identify the choice that best completes the statement or answers the question.

1. A weather balloon with a 2-meter diameter at ambient temperature holds 525 grams of helium. What type of electronic probe could be used to determine the pressure inside the balloon?
  - a. barometric
  - b. thermometric
  - c. calorimetric
  - d. spectrophotometric
2. A scientist observed changes in the gas pressure of one mole of a gas in a sealed chamber with a fixed volume. To identify the source of the changes, the scientist should check for variations in the
  - a. air pressure outside the chamber.
  - b. molecular formula of the gas.
  - c. temperature of the chamber.
  - d. isotopes of the gas.
3. Electrical fires cannot be safely put out by dousing them with water. However, fire extinguishers that spray solid carbon dioxide on the fire work very effectively. This method works because carbon dioxide
  - a. displaces the oxygen.
  - b. renders the fire's fuel non-flammable.
  - c. forms water vapor.
  - d. blows the fire out with strong wind currents.
4. In order to advance to the level of a theory, a hypothesis should be
  - a. obviously accepted by most people.
  - b. a fully functional experiment.
  - c. in alignment with past theories.
  - d. repeatedly confirmed by experimentation.
5. Matter is made of atoms that have positive centers of neutrons and protons surrounded by a cloud of negatively charged electrons. This statement is
  - a. a theory.
  - b. a hypothesis.
  - c. an inference.
  - d. an observation.
6. When a metal is heated in a flame, the flame has a distinctive color. This information was eventually extended to the study of stars because
  - a. the color spectra of stars indicate which elements are present.
  - b. a red shift in star color indicates stars are moving away.
  - c. star color indicates absolute distance.
  - d. it allows the observer to determine the size of stars.

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**Phillip John Buddle, J. S. Mackay**

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