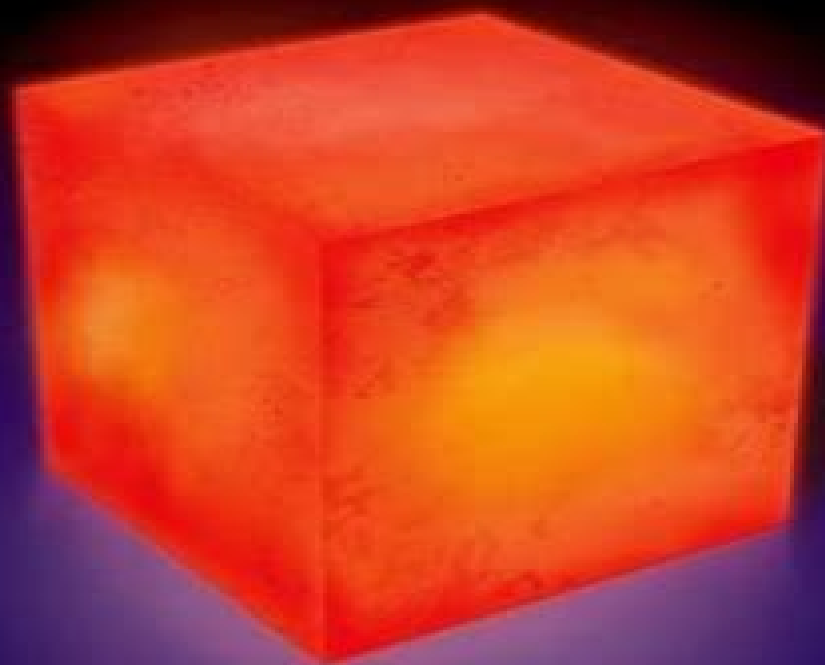


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Molecular Physical Chemistry for Engineering Applications Florin Emilian Daneş, Silvia Daneş, Valeria Petrescu, Eleonora-Mihaela Ungureanu, 2021-07-06 This textbook introduces the molecular side of physical chemistry. It offers students and practitioners a new approach to the subject by presenting numerous applications and solved problems that illustrate the concepts introduced for varied and complex technical situations. The book offers a balance between theory, tools, and practical applications. The text aims to be a practical manual for solving engineering problems in industries where processes depend on the chemical composition and physical properties of matter. The book is organized into three main topics: I the molecular structure of matter, II molecular models in thermodynamics, and III transport phenomena and mechanisms. Part I presents methods of analysis of the molecular behavior in a given system, while the following parts use these methods to study the equilibrium states of a material system and to analyze the processes that can take place when the system is in a state of non-equilibrium, in particular the transport phenomena. *Molecular Physical Chemistry for Engineering Applications* is designed for upper-level undergraduate and graduate courses in physical chemistry for engineers, applied physical chemistry, transport phenomena, colloidal chemistry, and transport transfer processes. The book will also be a valuable reference guide for engineers, technicians, and scientists working in industry. Offers modeling techniques and tools for solving exercises and practical cases. Provides solutions and conclusions so students can follow results more closely. Step-by-step problem solving enables students to understand how to approach complex issues.

Atkins' Physical Chemistry Peter Atkins, Julio de Paula, 2010 This volume features a greater emphasis on the molecular view of physical chemistry and a move away from classical thermodynamics. It offers greater explanation and support in mathematics, which remains an intrinsic part of physical chemistry.

What is Chemistry? Peter Atkins, 2013-08-22 Most people remember chemistry from their schooldays as a subject that was largely incomprehensible, fact-rich but understanding-poor, smelly and so far removed from the real world of events and pleasures that there seemed little point except for the most introverted in coming to terms with its grubby concepts, spells, recipes, and rules. Peter Atkins wants to change all that. In *What is Chemistry?* he encourages us to look at chemistry anew through a chemist's eyes to understand its central concepts and to see how it contributes not only towards our material comfort but also to human culture. Atkins shows how chemistry provides the infrastructure of our world through the chemical industry, the fuels of heating, power generation, and transport, as well as the fabrics of our clothing and furnishings. By considering the remarkable achievements that chemistry has made and examining its place between both physics and biology, Atkins presents a fascinating, clear, and rigorous exploration of the world of chemistry, its structure, core concepts, and exciting contributions to new cutting-edge technologies.

Atkins' Physical Chemistry Peter William Atkins, Julio De Paula, James Keeler, 2023 Revised edition of *Atkins Physical Chemistry* Peter Atkins, Julio de Paula, James Keeler, Eleventh edition 2018

The Chemistry Connection: From Atoms to Applications Dr. Sarika Arora, 2024-09-16

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Atkins Physical Chemistry 11th Edition Atkins, 2018-05

Elements of Physical Chemistry Peter William Atkins, Julio De Paula, 2017

This revision of the introductory textbook of physical chemistry has been designed to broaden its appeal, particularly to students with an interest in biological applications.

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The laws of thermodynamics drive everything that happens in the universe. From the sudden expansion of

a cloud of gas to the cooling of hot metal and from the unfurling of a leaf to the course of life itself everything is directed and constrained by four simple laws They establish fundamental concepts such as temperature and heat and reveal the arrow of time and even the nature of energy itself Peter Atkins powerful and compelling introduction explains what the laws are and how they work using accessible language and virtually no mathematics Guiding the reader from the Zeroth Law to the Third Law he introduces the fascinating concept of entropy and how it not only explains why your desk tends to get messier but also how its unstoppable rise constitutes the engine of the universe

Understanding the Failure of Materials and Structures David Jesson, 2024-09-11 Understanding the Failure of Materials and Structures introduces practical aspects of mechanical characterisation of materials and structures It gives those with little or no prior experience insight into the process of developing everyday products issues behind some high profile failures and tools to begin planning a programme of research Written in an easily accessible manner the work discusses fundamentals of the physical world highlighting the range of materials used and varied applications and offers a brief history of materials development It covers the role of materials structure in controlling materials properties and describes mechanical properties such as stress strain stiffness fracture and fatigue The book also features information on various modes of testing and strain measurement It provides some discussion on topics that go beyond well behaved test coupons with thoughts on biomechanics megastructures and testing for applications in extreme environments Finally it covers how materials fail and the future of physical testing With minimal theory and mathematics this work presents the fundamentals of mechanical characterisation of materials and structures in a manner accessible to the novice materials investigator and the layperson interested in the science behind materials engineered for use in common and advanced products

The Laws of Thermodynamics Peter Atkins, 2010-03-25 From the sudden expansion of a cloud of gas or the cooling of a hot metal to the unfolding of a thought in our minds and even the course of life itself everything is governed by the four Laws of Thermodynamics These laws specify the nature of energy and temperature and are soon revealed to reach out and define the arrow of time itself why things change and why death must come In this Very Short Introduction Peter Atkins explains the basis and deeper implications of each law highlighting their relevance in everyday examples Using the minimum of mathematics he introduces concepts such as entropy free energy and to the brink and beyond of the absolute zero temperature These are not merely abstract ideas they govern our lives In this concise and compelling introduction Atkins paints a lucid picture of the four elegant laws that between them drive the Universe ABOUT THE SERIES The Very Short Introductions series from Oxford University Press contains hundreds of titles in almost every subject area These pocket sized books are the perfect way to get ahead in a new subject quickly Our expert authors combine facts analysis perspective new ideas and enthusiasm to make interesting and challenging topics highly readable

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ample problems and tutorials throughout are much appreciated Tobin R Sosnick Professor and Chair of Biochemistry and Molecular Biology University of Chicago Presents both the concepts and equations associated with statistical thermodynamics in a unique way that is at visual intuitive and rigorous This approach will greatly benefit students at all levels Vijay S Pande Henry Dreyfus Professor of Chemistry Stanford University a masterful tour de force Barrick s rigor and scholarship come through in every chapter Rohit V Pappu Edwin H Murty Professor of Engineering Washington University in St Louis This book provides a comprehensive contemporary introduction to developing a quantitative understanding of how biological macromolecules behave using classical and statistical thermodynamics The author focuses on practical skills needed to apply the underlying equations in real life examples The text develops mechanistic models showing how they connect to thermodynamic observables presenting simulations of thermodynamic behavior and analyzing experimental data The reader is presented with plenty of exercises and problems to facilitate hands on learning through mathematical simulation Douglas E Barrick is a professor in the Department of Biophysics at Johns Hopkins University He earned his Ph D in biochemistry from Stanford University and a Ph D in biophysics and structural biology from the University of Oregon

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