

## 14.3

## HYDROGEN IONS AND ACIDITY

## Section Review

## Objectives

1. Calculate concentrations based on the pH of acids and bases and vice versa.
2. Calculate pH and concentrations based on the pH and  $K_a$  or  $K_b$  and vice versa.
3. Calculate the pH of weak acids and bases.

## Vocabulary

- |                   |                             |                  |
|-------------------|-----------------------------|------------------|
| 1. autoionization | 2. conjugate acid-base pair | 3. acid solution |
| 4. autoionization | 5. acid rain                | 6. acid rain     |
| 7. autoionization | 8. acid rain                | 9. acid          |

## Key Equation

- |                                     |                                     |
|-------------------------------------|-------------------------------------|
| 1. $\text{pH} = -\log [\text{H}^+]$ | 2. $\text{pH} = -\log [\text{H}^+]$ |
| 3. $\text{pH} = -\log [\text{H}^+]$ | 4. $\text{pH} = -\log [\text{H}^+]$ |

## Part A Completion

Write the equation and/or the chemical formula(s) of the species involved in the reaction. Write the equation and/or the chemical formula(s) of the species involved in the reaction.

1. Write the equation and/or the chemical formula(s) of the species involved in the reaction.
2. Write the equation and/or the chemical formula(s) of the species involved in the reaction.
3. Write the equation and/or the chemical formula(s) of the species involved in the reaction.
4. Write the equation and/or the chemical formula(s) of the species involved in the reaction.
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16. Write the equation and/or the chemical formula(s) of the species involved in the reaction.
17. Write the equation and/or the chemical formula(s) of the species involved in the reaction.
18. Write the equation and/or the chemical formula(s) of the species involved in the reaction.
19. Write the equation and/or the chemical formula(s) of the species involved in the reaction.
20. Write the equation and/or the chemical formula(s) of the species involved in the reaction.

1.  $\text{H}_2\text{O} \rightleftharpoons \text{H}^+ + \text{OH}^-$
2.  $\text{H}_2\text{O} \rightleftharpoons \text{H}^+ + \text{OH}^-$
3.  $\text{H}_2\text{O} \rightleftharpoons \text{H}^+ + \text{OH}^-$
4.  $\text{H}_2\text{O} \rightleftharpoons \text{H}^+ + \text{OH}^-$
5.  $\text{H}_2\text{O} \rightleftharpoons \text{H}^+ + \text{OH}^-$
6.  $\text{H}_2\text{O} \rightleftharpoons \text{H}^+ + \text{OH}^-$
7.  $\text{H}_2\text{O} \rightleftharpoons \text{H}^+ + \text{OH}^-$
8.  $\text{H}_2\text{O} \rightleftharpoons \text{H}^+ + \text{OH}^-$
9.  $\text{H}_2\text{O} \rightleftharpoons \text{H}^+ + \text{OH}^-$
10.  $\text{H}_2\text{O} \rightleftharpoons \text{H}^+ + \text{OH}^-$
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12.  $\text{H}_2\text{O} \rightleftharpoons \text{H}^+ + \text{OH}^-$
13.  $\text{H}_2\text{O} \rightleftharpoons \text{H}^+ + \text{OH}^-$
14.  $\text{H}_2\text{O} \rightleftharpoons \text{H}^+ + \text{OH}^-$
15.  $\text{H}_2\text{O} \rightleftharpoons \text{H}^+ + \text{OH}^-$
16.  $\text{H}_2\text{O} \rightleftharpoons \text{H}^+ + \text{OH}^-$
17.  $\text{H}_2\text{O} \rightleftharpoons \text{H}^+ + \text{OH}^-$
18.  $\text{H}_2\text{O} \rightleftharpoons \text{H}^+ + \text{OH}^-$
19.  $\text{H}_2\text{O} \rightleftharpoons \text{H}^+ + \text{OH}^-$
20.  $\text{H}_2\text{O} \rightleftharpoons \text{H}^+ + \text{OH}^-$

# 19 2 Hydrogen Ions And Acidity Section Review Answers

**Maurice Bleifeld**



## 19 2 Hydrogen Ions And Acidity Section Review Answers:

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treatment plant operator Information is provided on the importance of being an operator safety waste minimization physical  
chemical treatment process treatment of metal wastestreams and instrumentation **Russian Chemical Reviews** ,2006

## **19 2 Hydrogen Ions And Acidity Section Review Answers** Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has been apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**19 2 Hydrogen Ions And Acidity Section Review Answers**," compiled by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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