



COVALENT BONDING

Practice Problems

Do your answers to these very practice problems.

SECTION 8.1 MOLECULAR COMPOUNDS

1. Classify each of the following compounds as ionic or covalent.

- a. Na_2SO_4 b. CH_3NH_2 c. HCl d. H_2O e. H_2O_2 f. H_2O_2 g. H_2O_2 h. H_2O_2 i. H_2O_2 j. H_2O_2 k. H_2O_2 l. H_2O_2 m. H_2O_2 n. H_2O_2 o. H_2O_2 p. H_2O_2 q. H_2O_2 r. H_2O_2 s. H_2O_2 t. H_2O_2 u. H_2O_2 v. H_2O_2 w. H_2O_2 x. H_2O_2 y. H_2O_2 z. H_2O_2

2. Write the formula for the following compounds.

- a. H_2O b. H_2O_2 c. H_2O_2 d. H_2O_2 e. H_2O_2 f. H_2O_2 g. H_2O_2 h. H_2O_2 i. H_2O_2 j. H_2O_2 k. H_2O_2 l. H_2O_2 m. H_2O_2 n. H_2O_2 o. H_2O_2 p. H_2O_2 q. H_2O_2 r. H_2O_2 s. H_2O_2 t. H_2O_2 u. H_2O_2 v. H_2O_2 w. H_2O_2 x. H_2O_2 y. H_2O_2 z. H_2O_2

3. Write the formula for the following compounds.

a. H_2O b. H_2O_2 c. H_2O_2 d. H_2O_2 e. H_2O_2 f. H_2O_2 g. H_2O_2 h. H_2O_2 i. H_2O_2 j. H_2O_2 k. H_2O_2 l. H_2O_2 m. H_2O_2 n. H_2O_2 o. H_2O_2 p. H_2O_2 q. H_2O_2 r. H_2O_2 s. H_2O_2 t. H_2O_2 u. H_2O_2 v. H_2O_2 w. H_2O_2 x. H_2O_2 y. H_2O_2 z. H_2O_2

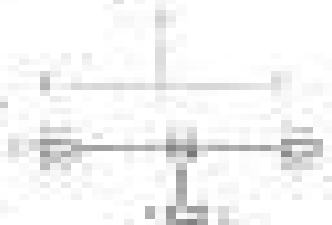
4. Write the formula for the following compounds.

SECTION 8.2 THE NATURE OF COVALENT BONDING

1. Draw the Lewis structure for the following compound.



2. Draw the Lewis structure for the following compound.



3. Draw the Lewis structure for the following compound.



4. Draw the Lewis structure for the following compound.

5. Draw the Lewis structure for the following compound.



SECTION 8.3 BONDING: THE ORBITAL

1. Write the formula for the following compound.

2. Write the formula for the following compound.

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Subrata Pal



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