

Name: _____

Period: _____

Atomic Structure



Atomic Number

Element Symbol

Element Name

Atomic Mass

The atomic number represents these two parts of the atom:

number of protons & number of electrons

You can calculate the number of neutrons by subtracting

Atomic Mass minus Atomic Number



Protons:

7

Electrons:

7

Neutrons:

7

Protons:

1

Electrons:

1

Neutrons:

0

Protons:

8

Electrons:

8

Neutrons:

8

BOHR DIAGRAMS

- Draw a circle in the center representing the nucleus.
- Label the number of protons and neutrons.
- Draw the electrons in their energy levels.
 1. The first energy level can hold 2 electrons
 2. The second energy level can hold 8 electrons
 3. The third energy level can hold 8 electrons

The electrons in the outermost energy level are called:

Valence Electrons

Element: SulfurNumber of Valence Electrons: 6

LEWIS DIAGRAMS

A lewis diagram shows only the symbol with the Valence electrons



Atomic Structure Worksheet Answer

Stewart C. Bushong

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

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 - The Rise of Digital Reading Atomic Structure Worksheet Answer
 - Advantages of eBooks Over Traditional Books
2. Identifying Atomic Structure Worksheet Answer
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Atomic Structure Worksheet Answer
 - User-Friendly Interface
4. Exploring eBook Recommendations from Atomic Structure Worksheet Answer

- Personalized Recommendations
- Atomic Structure Worksheet Answer User Reviews and Ratings
- Atomic Structure Worksheet Answer and Bestseller Lists
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 - Atomic Structure Worksheet Answer Public Domain eBooks
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 - Atomic Structure Worksheet Answer Budget-Friendly Options
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 - Atomic Structure Worksheet Answer Enhanced eBook Features
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 - Highlighting and Note-Taking Atomic Structure Worksheet Answer
 - Interactive Elements Atomic Structure Worksheet Answer
- 8. Staying Engaged with Atomic Structure Worksheet Answer
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 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Atomic Structure Worksheet Answer
- 9. Balancing eBooks and Physical Books Atomic Structure Worksheet Answer
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Atomic Structure Worksheet Answer
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Atomic Structure Worksheet Answer
 - Setting Reading Goals Atomic Structure Worksheet Answer
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Atomic Structure Worksheet Answer

- Fact-Checking eBook Content of Atomic Structure Worksheet Answer
- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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