

Atomic Theory & Atom Structure: Multiple Choice

1. Which one of the following is not one of the postulates of Dalton's atomic theory?

- AD Atoms are composed of protons, neutrons, and electrons.
- BD All atoms of a given element are identical; the atoms of different elements are different and have different properties.
- CD Atoms of an element are not changed into different types of atoms by chemical reactions.
- ED Atoms are neither created nor destroyed in chemical reactions.
- ED Compounds are formed when atoms of more than one element combine; a given compound always has the same relative number and kind of atoms.
- ED Each element is composed of extremely small particles called atoms.

ANS: A

2. Consider the following selected postulates of Dalton's atomic theory.

- CD Each element is composed of extremely small particles called atoms.
- CD Atoms are indivisible.
- CD Atoms of a given element are identical.
- CD Atoms of different elements are different and have different properties.

3. Which of the postulates is/are no longer considered valid?

- AD (D) and (E) BI (C) only CI (B) and (D) ED (B) only
- CD (B)

ANS: C

4. All atoms of a given element have the same _____.

- AD mass BI number of protons CD number of neutrons
- ED number of electrons and neutrons ED electrons

ANS: D

5. The nucleus of an atom does not contain _____.

- AD protons BI protons or neutrons CD neutrons
- ED subatomic particles ED electrons

ANS: B

6. The nucleus of an atom contains _____.

Atomic Structure Multiple Choice Answers

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Atomic Structure Multiple Choice Answers:

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Atomic Structure Multiple Choice Answers Book Review: Unveiling the Magic of Language

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