

# Atomic Structure



Atomic Number
Element Symbol
Element Name
Atomic Mass

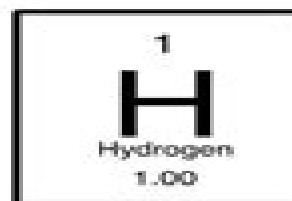
The atomic number represents these two parts of the atom:  
number of protons & number of electrons

You can calculate the number of neutrons by subtracting

Atomic Mass minus Atomic Number



Protons: 7  
Electrons: 7  
Neutrons: 7



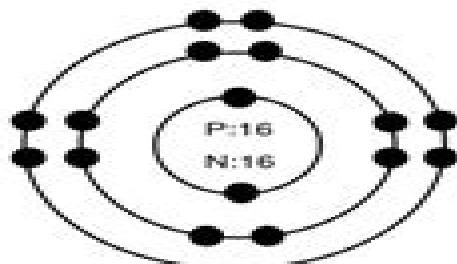
Protons: 1  
Electrons: 1  
Neutrons: 0



Protons: 8  
Electrons: 8  
Neutrons: 8

## BOHR DIAGRAMS

- Draw a circle in the center representing the nucleus.
- Label the number of protons and neutrons.
- Draw the electrons in their energy levels.
  1. The first energy level can hold 2 electrons
  2. The second energy level can hold 8 electrons
  3. The third energy level can hold 8 electrons



Element: Sulfur

Number of Valence Electrons: 6

The electrons in the outermost energy level are called:

Valence Electrons

## LEWIS DIAGRAMS

A Lewis diagram shows only the symbol with the Valence electrons



# Atomic Structure Worksheet Answer

**Richard Bailey**

## Atomic Structure Worksheet Answer:

[Analyze Your Fighting](#) Darin Waugh, 1999-01-01 The Analyze Your Fighting Method presented in this book shows martial artists how to use videotaped competition fights or sparring to truly understand their present skill and what it takes to become not just a great fighter but a superior fighter. The Analyze Your Fighting Method includes five levels of evaluation which are used to breakdown and study videotaped sparring and or competition matches from every technique that was thrown to how mental attitude affected performance. Then this book explains how to set fighting goals and plan workouts to achieve those goals. It also teaches martial artists how to use tactics to create winning strategies. To help in the creation of strategies this book includes a one of kind Tactics Catalog. This catalog lists over 100 tactics that fighters can incorporate into their strategies. This book is applicable to any type of martial arts system or competition including mixed martial arts.

[\*\*Exercises for Principles of Imaging Science and Protection\*\*](#) Michael A. Thompson, Steven B. Dowd, Janice D. Hall, Marian P. Hattaway, 1994 A smart collection of additional review questions and practise problems to further reinforce basic concepts. The laboratory manual section includes about 30 basic experiments which concentrate more on radiographic factors which affect the quality of the resulting radiograph. Readers with a weaker math science background will appreciate the remediation sections and self tests conclude each section to gauge the reader's weaknesses. [\*\*Teaching and Learning Online\*\*](#) Franklin S. Allaire, Jennifer E. Killham, 2023-01-10 Teaching and Learning Online Science for Elementary Grade Levels explores the challenges of teaching science virtually. It includes sections on frameworks, teacher journeys and lesson plans aligned with Next Generation Science Standards offering tips, resources and discussion questions for educators and students.

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