

Chapter 4

Atomic Structure

Class 8 - Dalal Simplified Middle School Chemistry Solutions

The atom - ancient views

Question 1

State the main postulates of - Dalton's atomic theory. Explain how the modern atomic theory contradicted Dalton's atomic theory.

Answer

Postulates of Dalton's atomic theory : Matter consists of particles called - atoms, which are indivisible & cannot be created or destroyed.

1. Atoms of the same element are alike but differ from atoms of all other elements.
2. Compounds or molecules are formed when atoms combine with other atoms in simple whole number ratios.
3. An atom is the smallest unit of matter which takes part in a chemical reaction.

Contradiction :

Modern atomic theory
Atoms are destructible
Atoms are divisible & consist of - protons, electrons & neutrons
Atoms of the same element may not be alike as seen in case of isotopes

Dalton's atomic theory
Atoms are indestructible
Atoms are indivisible & smallest particle of an element.
Atoms of the same element are alike in all respect.

Sub-atomic particles

Question 2

With reference to the discovery of the structure of an atom, explain in brief - William Crookes experiment for the discovery of cathode rays, followed by - J.J. Thomson's

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Arthur Erich Haas



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Table of Contents Atomic Structure Chapter 4

1. Understanding the eBook Atomic Structure Chapter 4
 - The Rise of Digital Reading Atomic Structure Chapter 4
 - Advantages of eBooks Over Traditional Books
2. Identifying Atomic Structure Chapter 4
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Atomic Structure Chapter 4
 - User-Friendly Interface
4. Exploring eBook Recommendations from Atomic Structure Chapter 4
 - Personalized Recommendations
 - Atomic Structure Chapter 4 User Reviews and Ratings
 - Atomic Structure Chapter 4 and Bestseller Lists

5. Accessing Atomic Structure Chapter 4 Free and Paid eBooks
 - Atomic Structure Chapter 4 Public Domain eBooks
 - Atomic Structure Chapter 4 eBook Subscription Services
 - Atomic Structure Chapter 4 Budget-Friendly Options
6. Navigating Atomic Structure Chapter 4 eBook Formats
 - ePub, PDF, MOBI, and More
 - Atomic Structure Chapter 4 Compatibility with Devices
 - Atomic Structure Chapter 4 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Atomic Structure Chapter 4
 - Highlighting and Note-Taking Atomic Structure Chapter 4
 - Interactive Elements Atomic Structure Chapter 4
8. Staying Engaged with Atomic Structure Chapter 4
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Atomic Structure Chapter 4
9. Balancing eBooks and Physical Books Atomic Structure Chapter 4
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Atomic Structure Chapter 4
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Atomic Structure Chapter 4
 - Setting Reading Goals Atomic Structure Chapter 4
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Atomic Structure Chapter 4
 - Fact-Checking eBook Content of Atomic Structure Chapter 4
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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