

Name: _____
Date/Period: _____

Atomic Number and Mass Number

Complete the following chart and answer the questions below.

Element Name	Atomic Number	Number of Protons	Number of Neutrons	Mass Number
carbon				12
	6		6	
hydrogen				1
		1		4
hydrogen			2	
nitrogen				14
			1	2
	92		146	
Potassium			20	
	11		12	
		47		108
Xenon			77	
			45	80
		24		52
			16	32
Gold				197
	76		114	

How are the atomic number and the number of protons related to each other?

How do the number of protons, number of neutrons, and the mass number relate to each other?

What is the one thing that determines the identity of an atom?

What sub atomic parts are equal to make a neutral atom?

Explain in your words how to calculate the mass of a sodium atom.

Draw and label three element symbols that are representative of the periodic table. It should have atomic #, mass #, name and element symbol.

Atomic Number Activity Answer

James L. Flowers

Atomic Number Activity Answer:

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