

# Atomic Theory Practice Test

\_\_\_\_\_ 1. If two atoms of Ca have a different number of electrons, which property (or properties) would be significantly different?

- a) Mass
- b) Charge
- c) Both A & B
- d) Neither A nor B

\_\_\_\_\_ 2. If two atoms of Na have a different number of neutrons which property (or properties) would be significantly different?

- a) Mass
- b) Charge
- c) Both A & B
- d) Neither A nor B

\_\_\_\_\_ 3. Which of the following ions will NOT have the same electron configuration as Ne?

- a)  $\text{Na}^+$
- b)  $\text{Al}^{3+}$
- c)  $\text{Ar}$
- d)  $\text{O}^{2-}$

\_\_\_\_\_ 4. Which of the following easily loses one outermost electron?

- a) Potassium
- b) Beryllium
- c) Iron
- d) Bromine

\_\_\_\_\_ 5. Which of the following would have the largest atomic radius?

- a) Seaborgium
- b) Einsteinium
- c) Indium
- d) Thallium

\_\_\_\_\_ 6. Which trend in the halogen family occurs with increasing atomic number?

- a) Ionization energies decrease
- b) Atomic radii decrease
- c) Electronegativities increase
- d) Tendency to gain electrons increases

\_\_\_\_\_ 7. Which of the following species would have 2 valence electrons?

- a) Boron
- b) Fluorine
- c) Oxygen ion
- d) Vanadium (III) ion

\_\_\_\_\_ 8. What sub-shell is especially stable when it is half-filled?

- a) s-subshell
- b) p-subshell
- c) d-subshell
- d) f-subshell

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**Carol Anne Neville**

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