

Atomic Theory & Atom Structure: Multiple Choice

1. Which one of the following is not one of the postulates of Dalton's atomic theory?
- AD Atoms are composed of protons, neutrons, and electrons.
 - BD All atoms of a given element are identical; the atoms of different elements are different and have different properties.
 - CD Atoms of an element are not changed into different types of atoms by chemical reactions.
 - ED Atoms are neither created nor destroyed in chemical reactions.
 - ED Compounds are formed when atoms of more than one element combine; a given compound always has the same relative number and kind of atoms.
 - ED Each element is composed of extremely small particles called atoms.

ANS: A

1. Consider the following selected postulates of Dalton's atomic theory.
- CD Each element is composed of extremely small particles called atoms.
 - CD Atoms are indivisible.
 - CD Atoms of a given element are identical.
 - CD Atoms of different elements are different and have different properties.

2. Which of the postulates is/are no longer considered valid?

- AD C and D
- BD C only
- CD C and D
- ED C and D only
- ED C and D

ANS: C

1. All atoms of a given element have the same _____.
- AD mass
 - BD number of protons
 - CD number of neutrons
 - ED number of electrons

ANS: D

1. The nucleus of an atom does not contain _____.
- AD protons
 - BD protons or neutrons
 - CD subatomic particles
 - ED electrons

ANS: B

1. The nucleus of an atom contains _____.

Atomic Structure Multiple Choice With Answers

Arshad Iqbal

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