



7. Which of the following is the atomic number of the atom represented by the diagram on the left?

- A. 22
- B. 25
- C. 47
- D. 2

8. What is the total electrical charge for the nucleus of an atom with 5 protons, 2 neutrons, and 10 electrons?

- A. +12
- B. -10
- C. +5
- D. -2

9. Which of the three following atoms has the greatest atomic mass?



1



2



3

- A. atom 1
- B. atom 2
- C. atom 3
- D. They all have equal atomic mass.

10. Which correctly describes the charges of the subatomic particles of an atom?

- A. electron/negative, proton/neutral, neutron/positive
- B. proton/positive, neutron/neutral, electron/negative
- C. neutron/negative, neutron/neutral, electron/positive
- D. neutron/negative, proton/positive, electron/neutral

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web satuan massa jenis yang sering digunakan adalah g cm<sup>3</sup> dimana 1 g cm<sup>3</sup> = 1000 kg m<sup>3</sup> walaupun zat itu beraneka ragam tetapi mempunyai beberapa sifat yang sama yaitu setiap zat menempati ruang dan mempunyai massa untuk zat padat dan zat cair mudah dibuktikan menempati ruang dan mempunyai massa

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web massa jenis yaitu perbandingan antara massa dengan volume suatu zat dan tidak dipengaruhi gaya gravitasi bumi berat

jenis suatu zat berbeda dengan lainnya zat yaitu sesuatu yang memiliki massa dan memerlukan ruang berdasarkan wujudnya zat dapat dibedakan menjadi zat padat zat cair dan zat gas

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web besarnya tekanan hidrostatik tidak bergantung pada bentuk bejana dan jumlah zat cair dalam bejana tetapi tergantung pada massa jenis zat cair percepatan gravitasi bumi dan kedalamannya secara matematis tekanan hidrostatik disuatu titik misal didasar balok diturunkan dari konsep tekanan

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