

Electron configuration of every element in the periodic table

#	Element	Electron configuration
1	Hydrogen	$1s^1$
2	Helium	$1s^2$
3	Lithium	$1s^2 2s^1$
4	Beryllium	$1s^2 2s^2$
5	Boron	$1s^2 2s^2 2p^1$
6	Carbon	$1s^2 2s^2 2p^2$
7	Nitrogen	$1s^2 2s^2 2p^3$
8	Oxygen	$1s^2 2s^2 2p^4$
9	Fluorine	$1s^2 2s^2 2p^5$
10	Neon	$1s^2 2s^2 2p^6$
11	Sodium	$1s^2 2s^2 2p^6 3s^1$
12	Magnesium	$1s^2 2s^2 2p^6 3s^2$
13	Aluminum	$1s^2 2s^2 2p^6 3s^2 3p^1$
14	Silicon	$1s^2 2s^2 2p^6 3s^2 3p^2$
15	Phosphorous	$1s^2 2s^2 2p^6 3s^2 3p^3$
16	Sulfur	$1s^2 2s^2 2p^6 3s^2 3p^4$
17	Chlorine	$1s^2 2s^2 2p^6 3s^2 3p^5$
18	Argon	$1s^2 2s^2 2p^6 3s^2 3p^6$
19	Potassium	$1s^2 2s^2 2p^6 3s^2 3p^6 4s^1$
20	Calcium	$1s^2 2s^2 2p^6 3s^2 3p^6 4s^2$
21	Scandium	$1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^1$
22	Titanium	$1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^2$
23	Vanadium	$1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^3$
24	Chromium*	$1s^2 2s^2 2p^6 3s^2 3p^6 4s^1 3d^5$
25	Manganese	$1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^5$
26	Iron	$1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^6$
27	Cobalt	$1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^7$
28	Nickel	$1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^8$
29	Copper *	$1s^2 2s^2 2p^6 3s^2 3p^6 4s^1 3d^{10}$
30	Zinc	$1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10}$
31	Gallium	$1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^1$
32	Germanium	$1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^2$
33	Arsenic	$1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^3$
34	Selenium	$1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^4$
35	Bromine	$1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^5$
36	Krypton	$1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6$
37	Rubidium	$1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^1$
38	Strontium	$1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^2$
39	Yttrium	$1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^2 4d^1$
40	Zirconium	$1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^2 4d^2$
41	Nobium	$1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^2 4d^3$
42	Molybdenum*	$1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^1 4d^5$
43	Technetium	$1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^2 4d^5$

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1. Understanding the eBook Atomic Structure Electron Configuration Answer Key
 - The Rise of Digital Reading Atomic Structure Electron Configuration Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Atomic Structure Electron Configuration Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Atomic Structure Electron Configuration Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Atomic Structure Electron Configuration Answer Key
 - Personalized Recommendations
 - Atomic Structure Electron Configuration Answer Key User Reviews and Ratings
 - Atomic Structure Electron Configuration Answer Key and Bestseller Lists
5. Accessing Atomic Structure Electron Configuration Answer Key Free and Paid eBooks
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 - Atomic Structure Electron Configuration Answer Key eBook Subscription Services

- Atomic Structure Electron Configuration Answer Key Budget-Friendly Options
- 6. Navigating Atomic Structure Electron Configuration Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Atomic Structure Electron Configuration Answer Key Compatibility with Devices
 - Atomic Structure Electron Configuration Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Atomic Structure Electron Configuration Answer Key
 - Highlighting and Note-Taking Atomic Structure Electron Configuration Answer Key
 - Interactive Elements Atomic Structure Electron Configuration Answer Key
- 8. Staying Engaged with Atomic Structure Electron Configuration Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Atomic Structure Electron Configuration Answer Key
- 9. Balancing eBooks and Physical Books Atomic Structure Electron Configuration Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Atomic Structure Electron Configuration Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Atomic Structure Electron Configuration Answer Key
 - Setting Reading Goals Atomic Structure Electron Configuration Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Atomic Structure Electron Configuration Answer Key
 - Fact-Checking eBook Content of Atomic Structure Electron Configuration Answer Key
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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