

Atomic Structure Worksheet

1. Name the three particles of the atom and their respective charges.

Particle	Charge type	Charge

2. The nucleus has a _____ charge and consist of _____ and _____.
3. Electrons are located in the _____ of an atom and carry _____ charge.
4. The number of protons in one atom of an element determines the atom's _____, and the number of electrons determines _____ of an element.
5. The atomic number tells you the number of _____ in one atom of an element. It also tells you the number of _____ in a neutral atom of that element. The atomic number gives the "identity" of an element as well as its location on the Periodic Table. No two different elements will have the _____ atomic number.
6. Give the symbol and number of protons in one atom of:
- Beryllium _____ Fluorine _____ Magnesium _____ Sulfur _____
Cobalt _____ Tin _____ Arsenic _____ Gold _____
7. Give the symbol and number of electrons in a neutral atom of:
- Potassium _____ Argon _____ Calcium _____
Chromium _____ Cadmium _____ Lead _____
8. Give the isotope symbol and number of neutrons in one atom of the following elements. Show your calculations.

Cesium - 133 _____

Iron - 54 _____

Ruthenium - 106 _____

Nickel - 58 _____

Polonium - 209 _____

Iodine - 135 _____

Indium - 115 _____

Phosphorous - 31 _____

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