

Atomic Mass Atomic Number Worksheet

Name: _____

Date: _____

Name of Element	Symbol	Atomic Number	Atomic Mass	Protons	Neutrons	Electrons
copper	Cu	29	64	29	35	29
tin	Sn	50	119	50	69	50
iodine	I	53	127	53	74	53
uranium	U	92	238	92	146	92
potassium	K	19	39	19	20	19
lithium	Li	3	7	3	4	3
oxygen	O	8	16	8	8	8
gold	Au	79	197	79	118	79
sulfur	S	16	32	16	16	16
silver	Ag	47	108	47	61	47
chromium	Cr	24	52	24	28	24
cobalt	Co	27	59	27	32	27
nickel	Ni	28	59	28	31	28
zinc	Zn	30	65	30	35	30
aluminum	Al	13	27	13	14	13
mercury	Hg	80	201	80	121	80
platinum	Pt	78	195	78	117	78
iron	Fe	26	56	26	30	26
hydrogen	H	1	1	1	0	1
helium	He	2	4	2	2	2
beryllium	Be	4	9	4	5	4
magnesium	Mg	12	24	12	12	12
carbon	C	6	12	6	6	6
silicon	Si	14	28	14	14	14
chlorine	Cl	17	35	17	18	17
bismuth	Bi	83	209	83	126	83
boron	B	5	11	5	6	5
calcium	Ca	20	40	20	20	20
manganese	Mn	25	55	25	30	25
lead	Pb	82	207	82	125	82
sodium	Na	11	23	11	12	11
fluorine	F	9	19	9	10	9
phosphorus	P	15	31	15	16	15

Atomic Mass And Atomic Number Answer Key

Henry Watts



Atomic Mass And Atomic Number Answer Key:

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Confusing Science Terms, Grades 5 - 8 Schyrlet Cameron, Carolyn Craig, 2010-08-06 Connect students in grades 5 and up with science using Confusing Science Terms This 80 page book helps students differentiate between confused word pairs or triples and perplexing science terminology The book includes terms from the areas of physical life earth and space science It encourages students to use a science vocabulary journal to construct their own meanings for confusing terms write sentences using the terms and create visual representations for them Students increase their knowledge and understanding of science concepts through vocabulary building while improving science literacy This book includes decoding activities and alternative methods of instruction such as hands on and small group activities games and journaling which allow for differentiated instruction The book supports National Science Education Standards

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Chemical News and Journal of Industrial Science, 1897 *The Chemical News and Journal of Industrial Science*, 1882

A Dictionary of chemistry and the allied branches of other sciences v. 3, 1882 Henry Watts, 1882

The Chemical News and Journal of Physical Science, 1897 Chemical Interactions, 2005 Introduction to Chemical Principles Howard Stephen Stoker, 1990 Introduction to Chemical Principles is a text for students who have had little to no previous instruction in chemistry or who had such instruction long enough ago that a thorough review is needed preface

1,000 Questions and Answers, 1997 Questions and answers

Engineer-in-training License Review Donald G. Newnan, 1995 *The Chemical News*, 1872

Chemical News and Journal of Industrial Science, 1925

McGraw-Hill's 3 Evolve Reach (HESI) A2 Practice Tests Kathy A. Zahler, 2013 McGraw Hill's 3 Evolve Reach HESI A2 practice tests helps you prepare for this important exam and achieve the high score you need to get into nursing school you desire Back cover

Chemistry Charles H. Corwin, 1994 The book focuses on the

concepts of chemistry and the applications that maintain and generate motivation for the subject of chemistry

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