

Answer Sheet

Part A: Answer Questions

- What two properties make up the mass? **1. volume and density**
- Describe the composition of the atom.
- What two forces make the atoms hold together? **1. electromagnetic**
- Describe the structure of the atom. **1. nucleus and electron cloud**
- What is the name of the atom? **1. hydrogen**

Part B: Answer Questions

- What is the name of the atom? **1. hydrogen**

1	1.000	1.000	1.000
2	1.000	1.000	1.000
3	1.000	1.000	1.000
4	1.000	1.000	1.000
5	1.000	1.000	1.000
6	1.000	1.000	1.000
7	1.000	1.000	1.000
8	1.000	1.000	1.000
9	1.000	1.000	1.000
10	1.000	1.000	1.000

- What is the name of the atom? **1. hydrogen**
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Element	Atomic Number	Atomic Mass	Atomic Number	Atomic Mass	Atomic Number
H	1	1.008	2	4.003	3
He	2	4.003	3	7.016	4
Li	3	7.016	4	9.012	5
Be	4	9.012	5	10.013	6
B	5	10.013	6	12.011	7
C	6	12.011	7	14.003	8
N	7	14.003	8	16.000	9
O	8	16.000	9	18.998	10
F	9	18.998	10	20.015	11
Ne	10	20.015	11	22.990	12
Na	11	22.990	12	24.003	13
Mg	12	24.003	13	26.982	14
Al	13	26.982	14	28.086	15
Si	14	28.086	15	30.974	16
P	15	30.974	16	32.065	17
S	16	32.065	17	35.453	18
Cl	17	35.453	18	39.962	19
Ar	18	39.962	19	40.078	20

These are the atomic numbers and atomic masses of the elements. The atomic number is the number of protons in the nucleus of an atom. The atomic mass is the sum of the protons and neutrons in the nucleus of an atom.

Atomic Basics Answer Key

Barbara Herlihy

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the author's name.

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