

# **APUSH/USHISTORY – MULTIPLE CHOICE REVIEW FOR MIDTERM**

1. Civilizations in the Americas before the arrival of Columbus
  - a. Were only small bands of hunters and gatherers
  - b. Were substantial cultures with superior technology compared to Europe
  - c. Included elaborate and sophisticated cultures that rivaled those of Europe
  - d. Were nomadic tribes that used horses for hunting
  - e. Had no written language, public buildings, or agriculture.
2. The Treaty of Tordesilas contained a provision to
  - a. Give Spain all rights to settle in the Americas
  - b. End Aztec domination in Mexico
  - c. End the rebellion of the Pueblos in New Mexico
  - d. Divide the Americas between Spain and Portugal to colonize
  - e. Allow England to colonize North America and Spain to colonize Central and South America
3. Which of the following were brought from America to Europe?
  - a. Horses
  - b. Potatoes
  - c. Bananas
  - d. Cattle
  - e. Sugar cane
4. A principle of the economic theory of mercantilism was that
  - a. Centralized control of the economy was essential to increasing a nation's wealth
  - b. Free trade and the market system would increase a country's productivity
  - c. The world's wealth increased as the population grew
  - d. Importing goods from other countries would conserve a nation's resources and make it wealthier
  - e. Government should not be involved in a nation's economy
5. Puritans in England were discontented with the Church of England because
  - a. Anglicans stressed spiritual matters more than worldly concerns
  - b. Puritans wanted to worship as they pleased in their own congregations, contrary to English law.
  - c. They wanted more elaborate rituals in the Church of England services
  - d. They wanted a more powerful role for bishops of the Church
  - e. Puritans wanted English Catholics to have the right to worship freely.
6. The first permanent English settlement in North America was
  - a. Roanoke Island
  - b. Quebec
  - c. New Amsterdam
  - d. Jamestown
  - e. Newfoundland
7. One of the problems in Jamestown that led to a high mortality rate in the early years was
  - a. Captain John Smith's lack of leadership and inability to impose order and discipline on the settlers
  - b. The settlement lay outside the territory of the helpful Powhatan tribe
  - c. The family groups were more susceptible to disease than single male settlers
  - d. The initiation of the Headright system, which promoted private property and migration of skilled workers
  - e. The cultivation of the staple crop tobacco for trade instead of grain.

# Apush Multiple Choice Section Practice

**Larry Krieger**



## **Apush Multiple Choice Section Practice:**

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spectator ions in the net chemical reaction the spectator ions are cancelled from both sides of the equation in other words  
spectate means watch the other ions react in an aqueous solution

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web these ions are called spectator ions because they do not participate in the actual reaction canceling the spectator ions  
gives the net ionic equation which shows only those species that participate in the chemical reaction  $\text{2Ag}^+ \text{aq}$   $\text{Cr}_2\text{O}_7^{2-} \text{aq}$   
 $\rightarrow \text{Ag}_2\text{Cr}_2\text{O}_7 \text{s}$  label 4 2 3

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web chemical equation in which all dissolved ionic reactants and products including spectator ions are explicitly represented by formulas for their dissociated ions molecular equation chemical equation in which all reactants and products are

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web jan 29 2020 the ionic form of this reaction is  $2\text{Na}^+\text{aq} + 2\text{Cl}^-\text{aq} + \text{Cu}^{2+}\text{aq} + \text{SO}_4^{2-}\text{aq} \rightarrow 2\text{Na}^+\text{aq} + \text{SO}_4^{2-}\text{aq} + \text{CuCl}_2\text{s}$  the sodium ions and sulfate ion are the spectator ions in this reaction they appear unchanged in both the product and reactant side of the equation

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web the six  $\text{NO}_3^-\text{aq}$  ions and the six  $\text{Na}^+\text{aq}$  ions that appear on both sides of the equation are spectator ions that can be canceled to give the net ionic equation  $3\text{Ba}^{2+}\text{aq} + 2\text{PO}_4^{3-}\text{aq} \rightarrow \text{Ba}_3(\text{PO}_4)_2\text{s}$

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web these spectator ions ions whose presence is required to maintain charge neutrality are neither chemically nor physically changed by the process and so they may be eliminated from the equation to yield a more succinct representation called a

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web the equation below is the final net ionic equation after elimination of the spectator ions which yields water  $\text{H}^+\text{aq} + \text{OH}^-\text{aq} \rightarrow \text{H}_2\text{O} + \text{l}$  the spectator ions in  $\text{KOH aq}$  and  $\text{HNO}_3\text{aq}$  are  $\text{K}^+\text{aq}$  and  $\text{NO}_3^-\text{aq}$  how to identify and list the

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web the ag aq and cl aq ions become agcl s but the na aq ions and the no 3 aq ions stay as na aq ions and no 3 aq ions these two ions are examples of spectator ions ions that do nothing in the overall course of a chemical reaction

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web aug 8 2022 a spectator ion is an ion that does not take part in the chemical reaction and is found in solution both before and after the reaction the net ionic equation is the chemical equation that shows only those elements compounds and ions that are directly involved in the chemical reaction

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