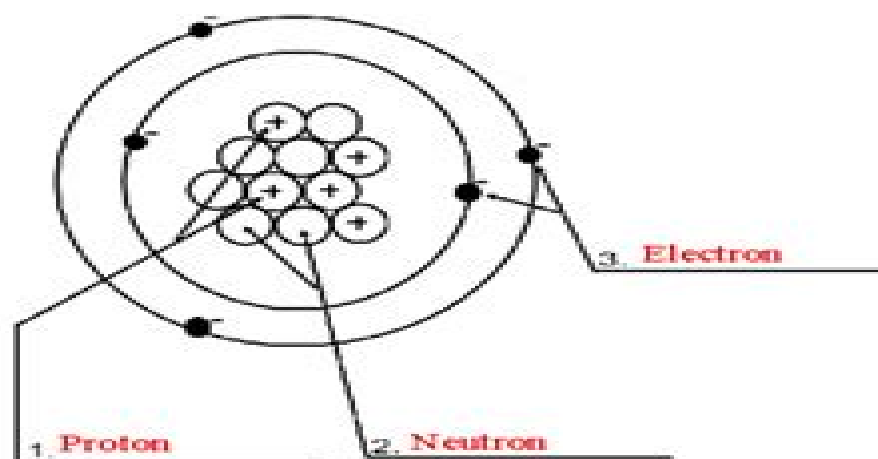


Atomic Structure Worksheet

Label the parts of an atom on the diagram below.



4. What type of charge does a proton have?

Positive (+1)

5. What type of charge does a neutron have?

Neutral (0)

6. What type of charge does an electron have?

Negative (-1)

7. Which two subatomic particles are located in the nucleus of an atom?

Protons and neutrons

8. If an atom has 35 protons in the nucleus, how many electrons will it have orbiting the nucleus?

35

9. What is the atomic number of the atom in the diagram above?

5

10. What is the atomic mass/mass number of the atom in the diagram above?

(11+11=22) *protons and neutrons added together) 22 is the atomic mass/mass number

11. How many protons are in the nucleus of an atom with an atomic number of 15?

15 *number of protons is the same as the atomic number

12. How many electrons are in the nucleus of an atom with an atomic number of 20?

20

13. How many neutrons are in the nucleus of an atom with an atomic number of 25?

(use Periodic Table for mass)
30 Neutrons. atomic mass: 54.938 ~ 55 atomic number: 25.....55-25=30

14. What is the mass number of an atom with 3 protons, 4 neutrons, and 3 electrons?

Mass number= 6.94. (Li-Lithium has an atomic number of 3 which means it has 3 protons.

15. How many neutrons are in the nucleus of an atom that has an atomic mass of 36 and an atomic number of 25?

Neutrons= 11 (atomic mass: 36 - 25 :atomic number)

Atomic Structure Worksheet Answer Key

Jianjun Gao



Atomic Structure Worksheet Answer Key:

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 - Advantages of eBooks Over Traditional Books
2. Identifying Atomic Structure Worksheet Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Atomic Structure Worksheet Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Atomic Structure Worksheet Answer Key
 - Personalized Recommendations
 - Atomic Structure Worksheet Answer Key User Reviews and Ratings
 - Atomic Structure Worksheet Answer Key and Bestseller Lists
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 - Highlighting and Note-Taking Atomic Structure Worksheet Answer Key
 - Interactive Elements Atomic Structure Worksheet Answer Key
- 8. Staying Engaged with Atomic Structure Worksheet Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Atomic Structure Worksheet Answer Key
- 9. Balancing eBooks and Physical Books Atomic Structure Worksheet Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Atomic Structure Worksheet Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Atomic Structure Worksheet Answer Key
 - Setting Reading Goals Atomic Structure Worksheet Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Atomic Structure Worksheet Answer Key
 - Fact-Checking eBook Content of Atomic Structure Worksheet Answer Key
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

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