

## PHYSICAL SCIENCE PRACTICE TEST WITH ANSWER KEY

1. A weak attraction between hydrogen atom in one molecule and an oxygen in other called \_\_\_\_\_.  
A. covalent bond  
B. Hydrogen bond  
C. Ionic bond  
D. Metallic bond
2. Which of the following substances is most likely a covalent compound?  
A.  $\text{CaCO}_3$   
B.  $\text{KNO}_3$   
C.  $\text{SO}_2$   
D.  $\text{NaOH}$
3. What is the element needed to bond with Na to create the most ionic bond?  
A. F  
B. Cl  
C. Br  
D. I
4. The boiling point of substance A is higher than that of substance B. What does this mean?  
A. Substance A is less viscous than substance B.  
B. Substance A evaporates faster than substance B.  
C. Substance B has greater surface tension than substance A.  
D. Substance B has weaker intermolecular forces than substance A.
5. What is the intermolecular force of attraction between the nucleus of one atom and the electrons of a nearby atom?  
A. Dipole-dipole force  
B. Ion-dipole force  
C. London dispersion force  
D. Van der Waals force
6. Which of the following has the greatest number of atoms?  
A.  $\text{Ca(OH)}_2$   
B.  $\text{Al}_2\text{O}_3$

# Answer Key To Physical Science And Studyworkbook

**Marcel A. Müller**



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## **Table of Contents Answer Key To Physical Science And Studyworkbook**

1. Understanding the eBook Answer Key To Physical Science And Studyworkbook
  - The Rise of Digital Reading Answer Key To Physical Science And Studyworkbook
  - Advantages of eBooks Over Traditional Books
2. Identifying Answer Key To Physical Science And Studyworkbook
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Answer Key To Physical Science And Studyworkbook
  - User-Friendly Interface
4. Exploring eBook Recommendations from Answer Key To Physical Science And Studyworkbook
  - Personalized Recommendations
  - Answer Key To Physical Science And Studyworkbook User Reviews and Ratings
  - Answer Key To Physical Science And Studyworkbook and Bestseller Lists
5. Accessing Answer Key To Physical Science And Studyworkbook Free and Paid eBooks
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  - Answer Key To Physical Science And Studyworkbook Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Answer Key To Physical Science And Studyworkbook
  - Highlighting and Note-Taking Answer Key To Physical Science And Studyworkbook
  - Interactive Elements Answer Key To Physical Science And Studyworkbook
- 8. Staying Engaged with Answer Key To Physical Science And Studyworkbook
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Answer Key To Physical Science And Studyworkbook
- 9. Balancing eBooks and Physical Books Answer Key To Physical Science And Studyworkbook
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Answer Key To Physical Science And Studyworkbook
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Answer Key To Physical Science And Studyworkbook
  - Setting Reading Goals Answer Key To Physical Science And Studyworkbook
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Answer Key To Physical Science And Studyworkbook
  - Fact-Checking eBook Content of Answer Key To Physical Science And Studyworkbook
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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