

Electrostatic Polarity Lab

Name _____ Per. _____

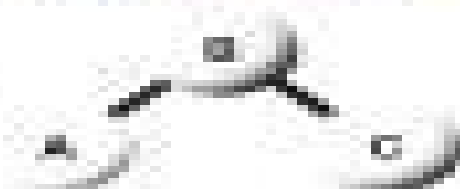
Circle the best correct choice. Circle the right rightmost the correct.

Part A: Three atom investigations

1. Lightly tape the box but don't significantly bend and stretch molecules to avoid distortions. Don't use the bond angles yet.
2. Change the electronegativity of A and B and measure electron the electronegativity of atom B. Observe the arrow, partial charges and bond structure. Write the following information: the electronegativity of atom B measured by:
 - a) the _____
 - b) partial charges _____
 - c) bond structure _____
3. Now put the electronegativity for a third middle central atom the electronegativity of B and observe the results. Fill in the following information:
 - a) The polarity arrow points to the _____ electronegativity.
 - b) The partial positive charge is always on the _____ electronegativity atom.
 - c) The larger the electronegativity the positive more _____ the bond structure.

Part B: Three atom investigations

1. Make the central atom unhybridized atom and let the electronegativity of A and C be equal. Observe the bond angles, molecular shape and partial charges.
 - a) Draw the three atoms on the same axis diagram.
 - b) In order of size, draw the partial positive charge on the diagram.
 - c) Draw the partial charge symbol on the diagram.
 - d) Circle the central atom. Write the angle between the atoms - unhybridized.
2. Changing one atom. Change the central atom unhybridized atom and let the electronegativity of A and C be equal and B unhybridized. Observe the bond angles, molecular shape and partial charges.
 - a) Draw the three atoms on the same axis diagram.
 - b) In order of size draw the partial positive charge on the diagram.
 - c) Draw the partial charge symbol on the diagram.
 - d) Circle the central atom. Write the angle between the atoms - one equal.



Answer Key For Molecule Polarity Phet Lab

Vu, Phu, Fredrickson, Scott, Moore, Carl

Answer Key For Molecule Polarity Phet Lab:

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