

# Chemistry Study Guide

| Periodic Table of the Elements |    |    |    |    |    |    |    |    |    |    |     |    |     |    |     |     |    |  |
|--------------------------------|----|----|----|----|----|----|----|----|----|----|-----|----|-----|----|-----|-----|----|--|
| H                              | Li | Be |    |    |    |    |    |    |    |    |     |    |     |    |     |     |    |  |
| Na                             | Mg |    |    |    |    |    |    |    |    |    |     |    |     |    |     |     |    |  |
| K                              | Ca | Sc | Ti | V  | Cr | Mn | Fe | Co | Ni | Cu | Zn  | Ga | Ge  | As | Se  | Br  | Kr |  |
| Rb                             | Sr | Y  | Zr | Nb | Mo | Tc | Ru | Rh | Pd | Ag | Cd  | In | Sn  | Sb | Te  | I   | Xe |  |
| Cs                             | Ba | Hf | Ta | W  | Re | Os | Ir | Pt | Au | Hg | Tl  | Pb | Bi  | Po | At  | Rn  |    |  |
| Fr                             | Ra | Rf | Db | Sg | Bh | Hs | Mt | Ds | Rg | Cn | Uut | Fl | Uup | Lv | Uus | Uuo |    |  |
|                                |    | La | Ce | Pr | Nd | Pm | Sm | Eu | Gd | Tb | Dy  | Ho | Er  | Tm | Yb  | Lu  |    |  |
|                                |    | Ac | Th | Pa | U  | Np | Pu | Am | Cm | Bk | Cf  | Es | Fm  | Md | No  | Lr  |    |  |

## Objectives:

To understand the structure of an atom, and the nature + properties of subatomic particles.



## Matter:

Anything that has mass and occupies space.

Examples: air, gases, liquids, solidz!

## Atoms:

Atoms are the smallest, indivisible particle.

- Outer part → shell/orbit/energy level → carries the electronz (e-)
- Inner part → nucleus → houses the protonz (p+) and neutronz (n)
- Outer & Inner part → together these are the subatomic particles.

## Subatomic Particle Table:

| Particle | Symbol         | Mass (amu) | Charge | Location    |
|----------|----------------|------------|--------|-------------|
| Proton   | p <sup>+</sup> | 1          | +1     | Nucleus     |
| Neutron  | n              | 1          | 0      | Nucleus     |
| Electron | e <sup>-</sup> | -0         | -1     | Orbit/Shell |

## Section 2 nuclear Chemistry Study Guide Key

**Ralph J. Fessenden,Joan S. Fessenden**

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