

15-2 Practice Problems

What is the molarity of the solution prepared when 14.3 g of phosphoric acid (H_3PO_4) is dissolved in sufficient water to prepare 2.00 L of solution?

$$\frac{14.3 \text{ g}}{97.99 \text{ g/mol}} \times \frac{1 \text{ mol}}{1 \text{ mol}} = 0.146 \text{ mol}$$

Phosphoric acid (molar mass 97.99) can be used to prepare 2.00 L of a 0.146 M solution of phosphoric acid.

$$\frac{14.3 \text{ g}}{97.99 \text{ g/mol}} \times \frac{1 \text{ mol}}{1 \text{ mol}} = 0.146 \text{ mol}$$

What is the molarity of the solution prepared when 14.3 g of phosphoric acid (H_3PO_4) is dissolved in sufficient water to prepare 2.00 L of solution?

$$\frac{14.3 \text{ g}}{97.99 \text{ g/mol}} \times \frac{1 \text{ mol}}{1 \text{ mol}} = 0.146 \text{ mol}$$

To prepare 2.00 L of a 0.146 M solution of phosphoric acid (H_3PO_4), how many grams of phosphoric acid must be dissolved?

$$\frac{2.00 \text{ L} \times 0.146 \text{ mol/L}}{1 \text{ mol}} \times 97.99 \text{ g/mol} = 28.6 \text{ g}$$

If 2.00 g of phosphoric acid (H_3PO_4) are dissolved in sufficient water to make 2.00 L of solution, what is the molarity of the solution?

$$\frac{2.00 \text{ g}}{97.99 \text{ g/mol}} \times \frac{1 \text{ mol}}{1 \text{ mol}} = 0.0204 \text{ mol}$$

$$\frac{0.0204 \text{ mol}}{2.00 \text{ L}} = 0.0102 \text{ M}$$

To make 2.00 L of a 0.146 M solution of phosphoric acid (H_3PO_4), how many grams of phosphoric acid must be dissolved?

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Michael Rajnik



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Problem-Solving Math Journals for Primary Students Mary Rosenberg, 2000 Activities designed to develop logical and mathematical thinking skills of primary school students

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In a digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**15 2 Practice Problems Answer Key**," compiled by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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