

SECOND TEST DECEMBER-2023

CLASS : 1 PUC

SUBJECT : CHEMISTRY (34)

Max. Marks : 35

TIME : 1 Hr. 30 Mins.

PART-A

I. Select the correct option from the given choices :

6x1=6

1) Relation between K_p and K_c is given by

a) $K_p = K_c (RT)^{\Delta n}$

b) $K_c = K_p (RT)^{\Delta n}$

c) $K_p = K_c / (RT)^{\Delta n}$

d) $K_p = (RT)^{\Delta n} / K_c$

2) At 298 K, $K_a \times K_b$ is

a) 10^{-7}

b) 10^{-14}

c) 10^{-10}

d) 10^{-7}

3) The process of addition of electrons is called

a) Reduction

b) Oxidation

c) Displacement reaction

d) Decomposition reaction

4) The oxidation number of an atom in the elemental state is

a) +1

b) -1

c) zero

d) $-\frac{1}{2}$

5) Soda lime is a mixture of



d) All of there

6) In benzene each carbon atom undergoes

a) sp^3 hybridizationb) sp^2 hybridizationc) sp hybridization

d) None of there

II. Fill in the blanks by choosing appropriate word from those given in brackets: 3x1=3

[$\text{C}_n\text{H}_{2n-2}$, C_nH_{2n} , Zero, HSO_4^-]7) Conjugate base of H_2SO_4 is _____.

8) The standard electrode potential of standard hydrogen electrode is _____.

9) The general formula of alkyne is _____.

PART-B

III. Answer any TWO of the following questions. Each question carries TWO marks:

2x2=4

10) What is heterogeneous equilibrium? Give an example.

11) What is conjugate acid-base pair? Illustrate with an example.

12) Explain the preparation of alkenes by dehydrohalogenation of alkyl halides.

13) Write the Newman's projection of ethane.

PART-C

IV. Answer any TWO of the following questions. Each question carries THREE marks:

2x3=6

14) Balance the following redox reaction by oxidation number method in acidic medium



15) Define oxidising agent. Give one example each for oxidising and reducing agents.

(P.T.O)

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Michael Rajnik



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